

# Electronic State Analysis of ZrB<sub>2</sub>, NbB<sub>2</sub>(0001) Surface Termination by Two-Dimensional Photoelectron Spectroscopy

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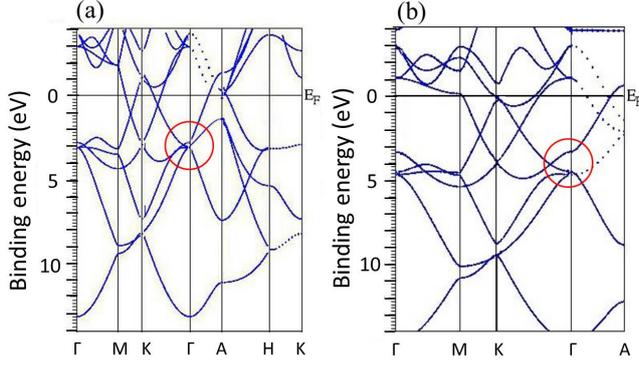
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## 1. Introduction

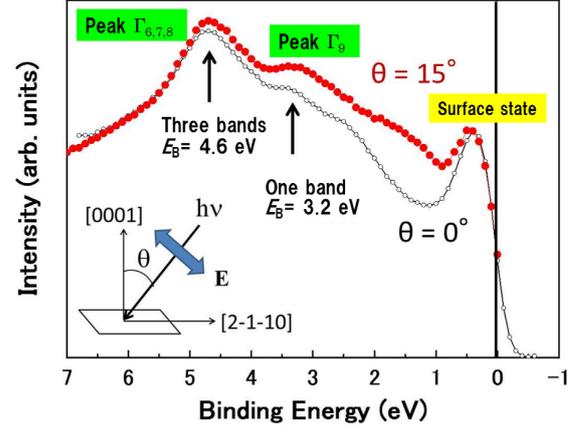
Zirconium diboride (ZrB<sub>2</sub>) and Niobium diboride (NbB<sub>2</sub>) are attractive materials for many applications. They have high melting points over 3000°C, high electric conductivities, and high corrosion resistance. Interestingly, ZrB<sub>2</sub> and NbB<sub>2</sub> have the same crystal structure as MgB<sub>2</sub>, which is an AlB<sub>2</sub>-type, where honeycomb boron (boraphene) sheets and close-packed metal layers are stacked alternately. However, the clean surface of ZrB<sub>2</sub>(0001) terminates with a Zr layer, whereas the clean NbB<sub>2</sub>(0001) surface terminates with a B layer [1]. The electronic structures of ZrB<sub>2</sub>(0001) and NbB<sub>2</sub>(0001) have been studied experimentally by ARUPS (angle-resolved ultraviolet photoelectron spectroscopy) [2] and theoretically [3]; however, the detailed electronic structures of these diborides have not been clarified yet. The band structures of ZrB<sub>2</sub> and NbB<sub>2</sub> are almost the same, but we can see a difference at  $\Gamma$ -point between them as indicated by a red circle in **Fig. 1**. In this study, we focused on this difference at  $\Gamma$ -point and investigated both electronic states including surface states in view of atomic orbitals constituting the electronic states by using two-dimensional photoelectron spectroscopy (2D-PES). We obtained some knowledge about the origin of this interesting difference of termination between ZrB<sub>2</sub>(0001) and NbB<sub>2</sub>(0001).

## 2. Experiment

The experiment was performed at BL-7 of the SR center at Ritsumeikan University [4]. A NbB<sub>2</sub>(0001) sample was heated up to 1000 °C at first for degassing in the preparation chamber. After recovery of the vacuum, the sample was heated up to 1000 °C for 3 to 8 min several times under the O<sub>2</sub> gas pressure of  $\sim 5 \times 10^{-4}$  Pa. After the sample was flash-annealed at 1500 °C for a few seconds several times, the surface quality was checked by low energy electron diffraction (LEED) and Auger electron spectroscopy. 2D-PES measurements of these



**Fig. 1** Band structures of (a) ZrB<sub>2</sub> and (b) NbB<sub>2</sub> calculated by WIEN2k.



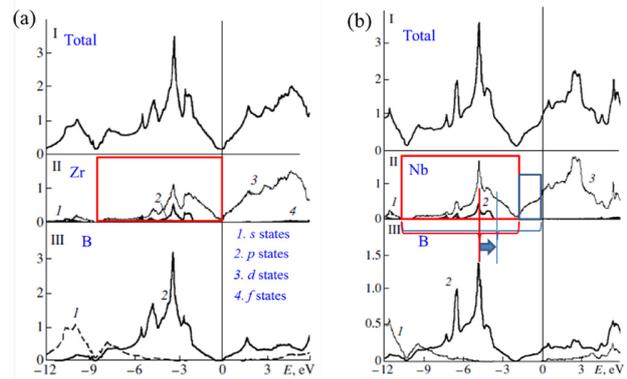
**Fig. 2** Polarization dependence of angle-resolved photoelectron spectra around the  $\bar{\Gamma}$ -point in NbB<sub>2</sub>(0001).

samples were performed at room temperature under an ultrahigh vacuum of  $\sim 1 \times 10^{-8}$  Pa by using a display-type spherical mirror analyzer (DIANA) [5]. The excitation light was a horizontally linearly-polarized synchrotron radiation (SR), and the photon energy was 40 eV. The total energy resolution was about 400 meV and the angular resolution was about  $1^\circ$ . The work function of DIANA was 4.2 eV.

### 3. Results and discussion

In **Fig.1**, the band structures of (a) ZrB<sub>2</sub> and (b) NbB<sub>2</sub> calculated by WIEN2k are shown. The band dispersions NbB<sub>2</sub> (b) are almost the same when the binding energies of those of ZrB<sub>2</sub> (a) are shifted downward by 1.5 eV. However there is a difference at the  $\bar{\Gamma}$  point indicated by red circles, that is ZrB<sub>2</sub> has four degenerated bands at the circle whereas NbB<sub>2</sub> has three degenerated bands and one separated band there. We measured two-dimensional photoelectron intensity angular distribution (PIAD) patterns with different incident angles to analyze atomic orbitals. We extracted the photoelectron intensities around  $\bar{\Gamma}$ -point from the PIADs measured at polar angles  $\theta = 0^\circ$  and  $15^\circ$ , and made the polarization dependent angle-resolved photoelectron spectra as seen in **Fig. 2**. We can see a surface state near the Fermi level and two peaks at the binding energies of  $E_B = 4.6$  eV and 3.2 eV. The observed peaks at  $E_B = 4.6$  eV and 3.2 eV correspond to the band structure of NbB<sub>2</sub> in **Fig. 1(b)** at  $E_B = 4.6$  eV (we call this peak as  $\Gamma_{6,7,8}$ ) and 3.2 eV (we call this peak as  $\Gamma_9$ ). The peak  $\Gamma_9$  is stronger in the spectra of  $\theta = 15^\circ$  than that of  $\theta = 0^\circ$ . This means that the possible atomic orbital constituting this band is  $p_z$  orbital. Hence this band is a bonding band between B layer and metal (Nb) layer [6]. We confirmed experimentally and theoretically that this inter-layer bonds became weaker in NbB<sub>2</sub> than in ZrB<sub>2</sub> because the energy of this band has higher energy than other three-degenerated bands as shown in **Fig. 1(b)**.

**Figures 3(a)** and **(b)** show the density-of-states (DOS) of ZrB<sub>2</sub> and NbB<sub>2</sub> [3]. Red rectangles show the DOS of Zr and Nb 4d orbitals, where the DOS of NbB<sub>2</sub> is 1.5 eV lower than that of ZrB<sub>2</sub> as seen in **Fig. 1**. Notice that Nb has one more 4d electron than Zr. The DOS of this electron appears near Fermi level of NbB<sub>2</sub> as shown by a blue rectangle in **Fig. 3(b)**. Then the center of gravity of Nb 4d DOS shifts upward a little as shown by a blue arrow in **Fig. 3(b)**. This indicates that the B layer is more stable than the metal layer in NbB<sub>2</sub>. As discussed above, the metal and B layers are more independent in NbB<sub>2</sub> than those in ZrB<sub>2</sub>. In summary the energy of the B layer is deeper than the metal layer in both NbB<sub>2</sub> and ZrB<sub>2</sub>, but it is further deeper in NbB<sub>2</sub> than in ZrB<sub>2</sub>, which can explain the reason why the B layer terminates on NbB<sub>2</sub>(0001) instead of the metal layer.



**Fig. 3** (I) Total and (II, III) local densities of valence (1) s states, (2) p states, (3) d states, and (4) f states of (II) metal and (III) boron sublattices of (a) ZrB<sub>2</sub> and (b) NbB<sub>2</sub> [3].

#### 4. Conclusion

To obtain a knowledge about the origin of the difference in the surface termination between ZrB<sub>2</sub> and NbB<sub>2</sub> (0001) clean surface in terms of the electronic structure, we measured two-dimensional photoelectron intensity angular distribution (PIAD) patterns by using a display-type spherical mirror analyzer (DIANA) at BL-7 of SR center, Ritsumeikan University. We measured the polarization dependence of the angle-resolved photoelectron spectra of NbB<sub>2</sub> at  $\bar{\Gamma}$  point at the binding energies around 4 eV. We found that the bonding between the B layer and the metal layer is weaker in NbB<sub>2</sub> than in ZrB<sub>2</sub>, which means the independency of each layer in NbB<sub>2</sub>. We concluded that the energy of B layer is deeper than the metal layer in both NbB<sub>2</sub> and ZrB<sub>2</sub>, but it is further deeper in NbB<sub>2</sub> than ZrB<sub>2</sub>, which can be the reason why the B layer terminates on NbB<sub>2</sub>(0001).

#### References

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