

Size change of Cu nanoparticles on rutile TiO₂(110) by oxygen exposure

Kazuma Ide¹, Toshitaka Aoki², Kei Mitsuhara¹, and Masaru Takizawa¹

1) Department of Physical Science, Faculty of Science and Engineering, Ritsumeikan University, 1-1-1 Noji-Higashi, Kusatsu 525-8577, Japan

2) The SR center, Ritsumeikan University, 1-1-1 Noji-Higashi, Kusatsu 525-8577, Japan

It is well known that Cu based catalysts are used in the reactions such as water gas-shift, methanol oxidation, CO oxidation, and so on. In spite of many efforts, however, the mechanism leading to strong catalytic activities of Cu nanoparticles is still a debatable issue. In the previous study [1], we have found the growth mode of Cu nanoparticles deposited on the rutile TiO₂(110) surfaces by medium energy ion scattering (MEIS). As a result, the two- (2D) dimensional islands grow initially up to Cu coverage of 0.5 ML and then the three- (3D) islands growth become dominant. In this work, we have measured the particle size of oxygen-exposed Cu nanoparticles by MEIS.

The experiments were performed at beamline 8 named SORIS set up at Ritsumeikan SR center. In the present MEIS analysis, we used 120 keV He⁺ ions and scattered He⁺ ions were energy-analyzed by a toroidal electrostatic analyzer (ESA) with an excellent energy resolution ($\Delta E/E = 9 \times 10^{-4}$, FWHM). We prepared the oxygen-rich surfaces (O-TiO₂) by exposure of O₂ onto the reduced TiO₂ surfaces, which were formed by sputtering with 0.75 keV Ar⁺ followed by annealing at 870 K for 10 min in UHV. Cu was deposited on the clean surfaces at RT with a Kundsens cell under UHV condition.

Fig. 1 (a) and (b) indicate the MEIS spectra observed for Cu(0.7 ML)/O-TiO₂(110) at an incident angle of 45° and emergent angle of 45° and 70°, respectively. The observed MEIS spectra are reproduced well assuming the following fitting parameters: d (diameter) = 3.1 nm, h (height) = 1.0 nm and δ (standard deviation in size) = 10 %. Interestingly, the average size of Cu nanoparticles has decreased and standard deviation has increased ($d = 2.5$ nm, $h = 0.95$ nm, $\delta = 32$ %) after O₂ exposure of 100 L (Fig. 2). Such size changes have also observed by J. Zhou *et. al.* [2]. The decrease in size of this Cu nanoparticles is probably come from that the adsorption of oxygen appears to weaken the Cu-Cu bond.

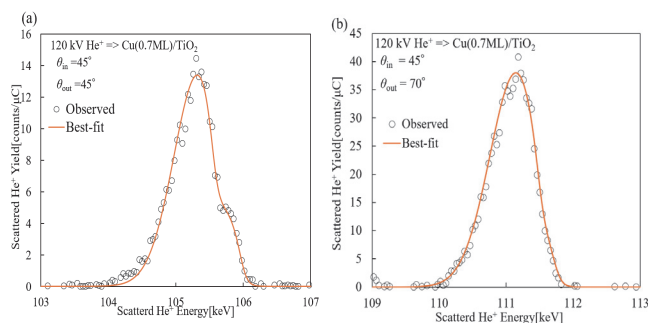


Fig. 1 MEIS spectra observed for 120 keV He⁺ ions scattered from Cu(0.7 ML)/O-TiO₂(110) at an incident angle of 45° and emergent angle of 45° (a) and 70° (b), respectively.

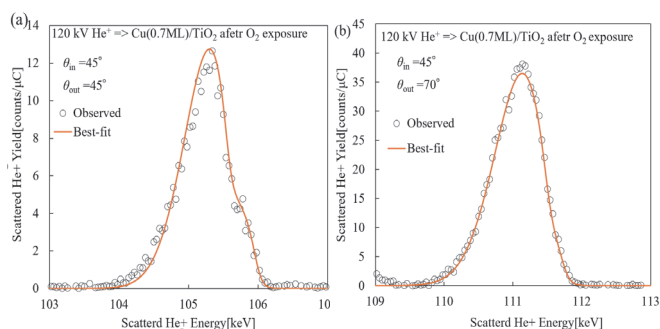


Fig. 2 MEIS spectra observed for 120 keV He⁺ ions scattered from Cu(0.7 ML)/O-TiO₂(110) after O₂ exposure of 100 L at an incident angle of 45° and emergent angle of 45° (a) and 70° (b), respectively.

References

- [1] T. Aoki, K. Mitsuhara, and M. Takizawa, *e-J. Surf. Sci.* **2018**, *16*, 225.
- [2] J. Zhou, Y. C. Kang, and D. A. Chen, *J. Phys. Chem. B* **2003**, *107*, 6664-6667.