

Li K-Edge XANES Spectra of Lithium-Doped Fullerenes and Lithium Borate Glasses

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Li K-edge XANES spectra for lithium doped fullerenes, Li_xC_{60} and Li_xC_{70} ($x = 14 \pm 2$), and for lithium borate glasses, $x\text{Li}_2\text{O}-(100-x)\text{B}_2\text{O}_3$, were measured using a beamline BL-2 of the SR Center at Ritsumeikan University [1]. All samples were powdered to collect their XANES spectra in the total electron yield mode. The vacuum level in the sample chamber was higher than 1.0×10^{-5} Pa during measurements. The lithium-doped fullerenes were synthesized at the laboratory of one of the authors (M. K.). For preparing lithium borate glasses, chemicals were melted at 1100-1200 and then quenched.

Figure 1 shows Li K-edge XANES spectra of Li_xC_{60} and Li_xC_{70} ($x = 14 \pm 2$) and reference samples. Spectra of **Li_xC₆₀ C 1** and **Li_xC₆₀ C 2** show the first and second runs of the measurements of a sample **Li_xC₆₀ C** under Ar atmosphere, respectively. Similar measurement was done for a different sample, **Li_xC₆₀ U 1**. A spectrum of **Li_xC₆₀ U 2** is for the sample partially exposed to the air. A spectrum of **Li_xC₇₀ E 1** is for a sample of Li_xC_{70} measured under Ar atmosphere, and that of **Li_xC₇₀ E air** is for the sample exposed to the air.

Broad and somewhat split absorptions appear around 57-69 eV in the spectra of **Li_xC₆₀ C 1**, **Li_xC₆₀ C 2**, and **Li_xC₆₀ U 1**, the whole features of which are similar to those of the reference samples such as Li_3N , but quite different from those of lithium halides like LiF, where the lithium atoms are in the isolated ionic states and show sharp absorption peaks due to the core exciton [2,3]. This indicates that lithium atoms in lithium-doped fullerenes do interact with C_{60} , like interacting Li and N atoms in Li_3N .

Compared to these absorptions, the corresponding broad absorption in **Li_xC₇₀ E 1** is somewhat shifted to higher energies, meaning that the Li- C_{70} interaction is different from the Li- C_{60} one.

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When exposed to the air, the spectral features (Li_xC_{60} U 2 and Li_xC_{70} E air) become similar, though the positions are a little in low energy, to those of the reference samples of Li_2CO_3 and Li_2O .

Figure 2 shows Li *K*-edge XANES spectra of lithium borate glasses $x\text{Li}_2\text{O}-(100-x)\text{B}_2\text{O}_3$ together with the reference samples. The whole spectral features of $x\text{Li}_2\text{O}-(100-x)\text{B}_2\text{O}_3$ ($x = 20-45$) resemble those of Li_2CO_3 and Li_2O . However, the spectra are rather broad with a relatively strong rising up at the low-energy side of the absorption. This may be indeed characteristic of the lithium atoms in amorphous glassy materials unlike crystalline ones.

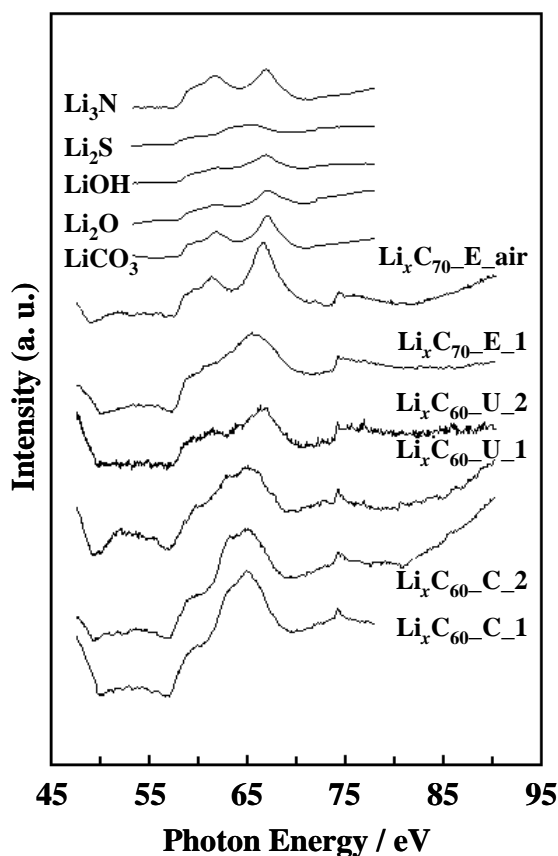


Figure 1. Li *K*-edge XANES spectra of Li_xC_{60} and Li_xC_{70} ($x = 14 \pm 2$) and reference samples.

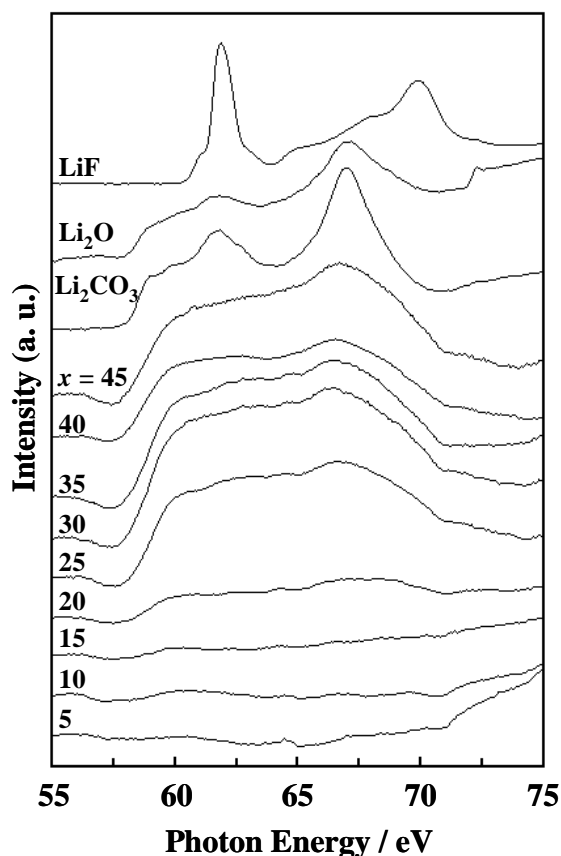


Figure 2. Li *K*-edge XANES spectra of lithium borate glasses $x\text{Li}_2\text{O}-(100-x)\text{B}_2\text{O}_3$ and reference samples.

References

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