

Electronic State Analysis of ZrB₂, NbB₂(0001) Surface Termination by Two-Dimensional Photoelectron Spectroscopy

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1. Introduction

Zirconium diboride (ZrB₂) and Niobium diboride (NbB₂) are attractive materials for many applications. They have high melting points over 3000°C, high electric conductivities, and high corrosion resistance. Interestingly, ZrB₂ and NbB₂ have the same crystal structure as MgB₂, which is an AlB₂-type, where honeycomb boron (boraphene) sheets and close-packed metal layers are stacked alternately. However, the clean surface of ZrB₂(0001) terminates with a Zr layer, whereas the clean NbB₂(0001) surface terminates with a B layer [1]. The electronic structures of ZrB₂(0001) and NbB₂(0001) have been studied experimentally by ARUPS (angle-resolved ultraviolet photoelectron spectroscopy) [2] and theoretically [3]; however, the detailed electronic structures of these diborides have not been clarified yet. The band structures of ZrB₂ and NbB₂ are almost the same, but we can see a difference at Γ -point between them as indicated by a red circle in **Fig. 1**. In this study, we focused on this difference at Γ -point and investigated both electronic states including surface states in view of atomic orbitals constituting the electronic states by using two-dimensional photoelectron spectroscopy (2D-PES). We obtained some knowledge about the origin of this interesting difference of termination between ZrB₂(0001) and NbB₂(0001).

2. Experiment

The experiment was performed at BL-7 of the SR center at Ritsumeikan University [4]. A NbB₂(0001) sample was heated up to 1000 °C at first for degassing in the preparation chamber. After recovery of the vacuum, the sample was heated up to 1000 °C for 3 to 8 min several times under the O₂ gas pressure of $\sim 5 \times 10^{-4}$ Pa. After the sample was flash-annealed at 1500 °C for a few seconds several times, the surface quality was checked by low energy electron diffraction (LEED) and Auger electron spectroscopy. 2D-PES measurements of these

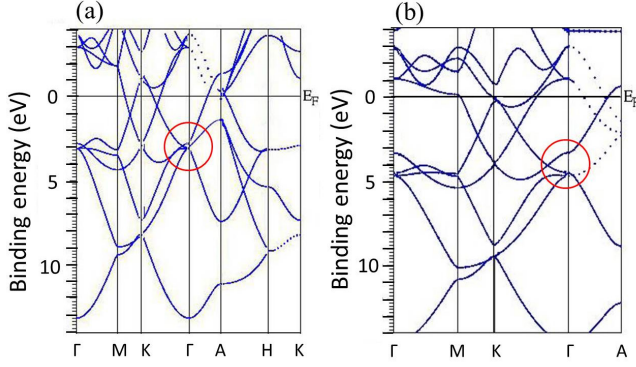


Fig. 1 Band structures of (a) ZrB₂ and (b) NbB₂ calculated by WIEN2k.

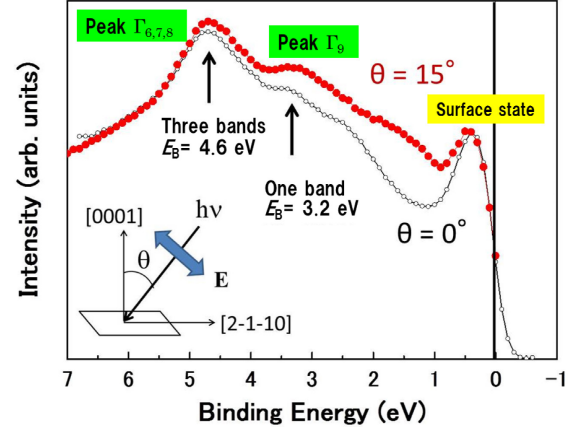


Fig. 2 Polarization dependence of angle-resolved photoelectron spectra around the $\bar{\Gamma}$ -point in NbB₂(0001).

samples were performed at room temperature under an ultrahigh vacuum of $\sim 1 \times 10^{-8}$ Pa by using a display-type spherical mirror analyzer (DIANA) [5]. The excitation light was a horizontally linearly-polarized synchrotron radiation (SR), and the photon energy was 40 eV. The total energy resolution was about 400 meV and the angular resolution was about 1° . The work function of DIANA was 4.2 eV.

3. Results and discussion

In **Fig.1**, the band structures of (a) ZrB₂ and (b) NbB₂ calculated by WIEN2k are shown. The band dispersions NbB₂ (b) are almost the same when the binding energies of those of ZrB₂ (a) are shifted downward by 1.5 eV. However there is a difference at the $\bar{\Gamma}$ point indicated by red circles, that is ZrB₂ has four degenerated bands at the circle whereas NbB₂ has three degenerated bands and one separated band there. We measured two-dimensional photoelectron intensity angular distribution (PIAD) patterns with different incident angles to analyze atomic orbitals. We extracted the photoelectron intensities around $\bar{\Gamma}$ -point from the PIADs measured at polar angles $\theta = 0^\circ$ and 15° , and made the polarization dependent angle-resolved photoelectron spectra as seen in **Fig. 2**. We can see a surface state near the Fermi level and two peaks at the binding energies of $E_B = 4.6$ eV and 3.2 eV. The observed peaks at $E_B = 4.6$ eV and 3.2 eV correspond to the band structure of NbB₂ in **Fig. 1(b)** at $E_B = 4.6$ eV (we call this peak as $\Gamma_{6,7,8}$) and 3.2 eV (we call this peak as Γ_9). The peak Γ_9 is stronger in the spectra of $\theta = 15^\circ$ than that of $\theta = 0^\circ$. This means that the possible atomic orbital constituting this band is p_z orbital. Hence this band is a bonding band between B layer and metal (Nb) layer [6]. We confirmed experimentally and theoretically that this inter-layer bonds became weaker in NbB₂ than in ZrB₂ because the energy of this band has higher energy than other three-degenerated bands as shown in **Fig. 1(b)**.

Figures 3(a) and **(b)** show the density-of-states (DOS) of ZrB₂ and NbB₂ [3]. Red rectangles show the DOS of Zr and Nb 4d orbitals, where the DOS of NbB₂ is 1.5 eV lower than that of ZrB₂ as seen in **Fig. 1**. Notice that Nb has one more 4d electron than Zr. The DOS of this electron appears near Fermi level of NbB₂ as shown by a blue rectangle in **Fig. 3(b)**. Then the center of gravity of Nb 4d DOS shifts upward a little as shown by a blue arrow in **Fig. 3(b)**. This indicates that the B layer is more stable than the metal layer in NbB₂. As discussed above, the metal and B layers are more independent in NbB₂ than those in ZrB₂. In summary the energy of the B layer is deeper than the metal layer in both NbB₂ and ZrB₂, but it is further deeper in NbB₂ than in ZrB₂, which can explain the reason why the B layer terminates on NbB₂(0001) instead of the metal layer.

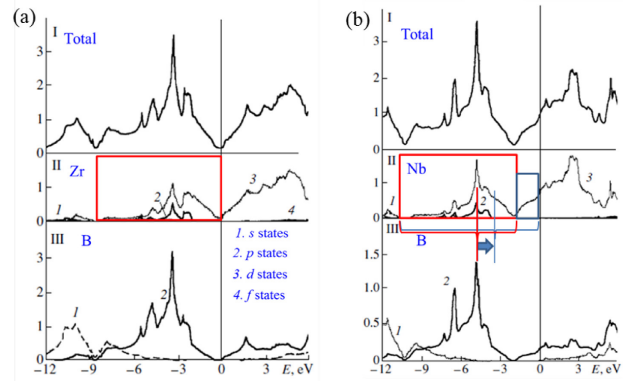


Fig. 3 (I) Total and (II, III) local densities of valence (1) s states, (2) p states, (3) d states, and (4) f states of (II) metal and (III) boron sublattices of (a) ZrB₂ and (b) NbB₂ [3].

4. Conclusion

To obtain a knowledge about the origin of the difference in the surface termination between ZrB₂ and NbB₂ (0001) clean surface in terms of the electronic structure, we measured two-dimensional photoelectron intensity angular distribution (PIAD) patterns by using a display-type spherical mirror analyzer (DIANA) at BL-7 of SR center, Ritsumeikan University. We measured the polarization dependence of the angle-resolved photoelectron spectra of NbB₂ at $\bar{\Gamma}$ point at the binding energies around 4 eV. We found that the bonding between the B layer and the metal layer is weaker in NbB₂ than in ZrB₂, which means the independency of each layer in NbB₂. We concluded that the energy of B layer is deeper than the metal layer in both NbB₂ and ZrB₂, but it is further deeper in NbB₂ than ZrB₂, which can be the reason why the B layer terminates on NbB₂(0001).

References

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