

# Soft X-ray core-level absorption study of Ni-based trinuclear complexes [Ni{Rh(ap<sub>t</sub>)<sub>3</sub>}<sub>2</sub>](NO<sub>3</sub>)<sub>n</sub> (n = 2, 3, 4)

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The Ni-Rh trinuclear complexes with 3-aminopropanethiolate (apt) [Ni{Rh(ap<sub>t</sub>)<sub>3</sub>}<sub>2</sub>](NO<sub>3</sub>)<sub>n</sub> (n = 2, 3, 4) have been successfully synthesized as a valence interconversion system [Fig. 1(a)] [1]. The nickel ions are in an octahedral coordination with six sulfur ions. The chemical analysis indicates that the Ni ions in [Ni{Rh(ap<sub>t</sub>)<sub>3</sub>}<sub>2</sub>](NO<sub>3</sub>)<sub>n</sub> are stable as a HS-Ni<sup>2+</sup> (d<sup>8</sup>) state for n = 2, a LS-Ni<sup>3+</sup> (d<sup>7</sup>) state for n = 3, and a LS-Ni<sup>4+</sup> (d<sup>6</sup>) state for n = 4, respectively.

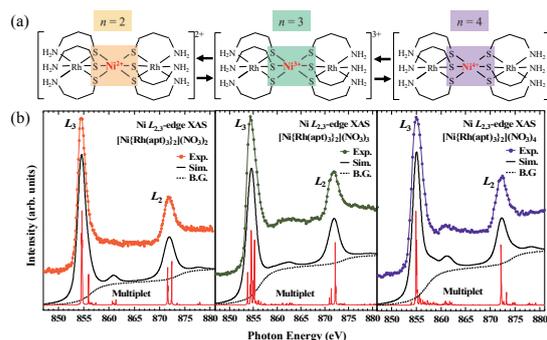
The importance of hybridization effects between the transition-metal (TM) and sulfur ions in inorganic sulfides for their physical properties have conventionally been recognized because of the fact that the radial distribution of S 3p orbital is larger than that of O and N 2p orbitals. Actually, the hybridization effects in NiS and NiS<sub>2</sub> have been reported on the basis of Ni 2p core-level photoemission and Ni L<sub>2,3</sub>-edge x-ray absorption (XAS) [2,3]. However, the direct verifications of the local Ni 3d electronic states in [Ni{Rh(ap<sub>t</sub>)<sub>3</sub>}<sub>2</sub>](NO<sub>3</sub>)<sub>n</sub> are still lacking. We have thus examined the 3d electronic states of [Ni{Rh(ap<sub>t</sub>)<sub>3</sub>}<sub>2</sub>](NO<sub>3</sub>)<sub>n</sub> probed by the Ni L<sub>2,3</sub>-edge XAS.

The Ni L<sub>2,3</sub>-edge XAS measurements were carried out at BL-11 of Synchrotron Radiation Center in Ritsumeikan University, Japan. The Ni L<sub>2,3</sub>-edge XAS spectra were taken in the partial electron yield (PEY) with a photon energy resolution of about 600 meV. In the PEY mode, we applied the voltage of 550 V to the Au mesh, installed in the front of microchannel plate (MCP), for the Ni L<sub>2,3</sub>-edge measurements in order to suppress a strong background caused by the C, N and O K-edge absorptions in the spectra [4].

We have also performed spectral simulations of a [NiS<sub>6</sub>]<sup>12-n</sup> configuration-interaction cluster model under the O<sub>h</sub> symmetry by using the XTLS 9.0 program [5]. We have reproduced the Ni L<sub>2,3</sub>-edge XAS spectra by optimizing the charge transfer energy from S 3p to Ni 3d orbitals: Δ and the hybridization strength described as a Slater-Coster parameter: (pdσ).

Figure 1(b) shows the Ni L<sub>2,3</sub>-edge XAS spectra

of [Ni{Rh(ap<sub>t</sub>)<sub>3</sub>}<sub>2</sub>](NO<sub>3</sub>)<sub>n</sub>. The satellite structure, which reflects the explicit hybridization effects between the Ni 3d and S 3p orbitals, has been observed at higher photon energy side of ~ 6 eV in all complexes. Simulated XAS spectra by the cluster model are also shown in Fig. 1(b), by which the hybridization parameters in Table 1 have been optimized. We have found that a so-called negative charge-transfer-energy (Δ < 0) state is realized for n = 4, in which the total energy of the d<sup>7</sup>L̄ state is lower than that of the d<sup>6</sup> state (L̄ stands for a hole in the sulfur sites).



**Fig. 1** (a) Molecular structure of [Ni{Rh(ap<sub>t</sub>)<sub>3</sub>}<sub>2</sub>](NO<sub>3</sub>)<sub>n</sub>. (b) Ni L<sub>2,3</sub>-edge XAS spectra and spectral simulations of [Ni{Rh(ap<sub>t</sub>)<sub>3</sub>}<sub>2</sub>](NO<sub>3</sub>)<sub>n</sub>.

**Table 1** Hybridization parameters obtained by the cluster-model calculations.

	n = 2	n = 3	n = 4
Δ	3.5 eV	1.0 eV	-4.0 eV
(pdσ)	1.2 eV	1.8 eV	0.9 eV

## References

- [1] M. Kouno *et al.*, *Angew. Chem. Int. Ed.* **56**, 13762 (2017).
- [2] A. E. Bocquet *et al.*, *Phys. Rev. B* **46**, 3771 (1992).
- [3] M. Nakamura *et al.*, *Phys. Rev. B* **48**, 16942 (1993).
- [4] K. Yamagami *et al.*, *J. Phys. Soc. Jpn.* **86**, 074801 (2017).
- [5] A. Tanaka, T. Jo, *J. Phys. Soc. Jpn.* **63**, 2788 (1994).