

Electronic State Analysis of Carbon Material on Optics Produced by SR Irradiation

Masaru Takizawa

Department of Physical Sciences, Faculty of Science and Engineering, Ritsumeikan University, 1-1-1 Noji-Higashi, Kusatsu 525-8577, Japan

The beamlines at synchrotron radiation (SR) facilities consist of mainly mirrors and gratings (optics) in order to produce desired photon. Although these optics are installed in ultrahigh vacuum (UHV) chamber ($\sim 10^{-8}$ Pa), the surfaces irradiated with SR are covered with carbon contamination due to the residual gases such as hydrocarbon. Since the carbon contamination causes the difficulty to measure the C *K*-edge near-edge X-ray absorption fine structure, various procedures, e.g., ultraviolet (UV)/ozone cleaning, to remove the carbon contamination have been reported [1]. The carbon contamination is considered to be graphite-like one, deduced from the photon intensity spectra after the contaminated optics [1]. However, it is very difficult to directly investigate the electronic structures of the contaminated optics surface. In this study, we have prepared the SR irradiated Au-coated Si samples and investigated the electronic states during UV/ozone cleaning.

The experiments were performed at the linearly polarized soft X-ray beamline BL-7 of SR center, Ritsumeikan University. Au-coated Si samples were introduced into the UHV beamline chamber and irradiated by SR. Then, the samples were transferred to the X-ray photoelectron spectroscopy (XPS) chamber immediately. For the UV/ozone cleaning, the samples were irradiated by UV under the oxygen (O_2) pressure of ~ 100 Pa. After the UV/ozone cleaning, the chamber was evacuated to $\sim 10^{-6}$ Pa and the samples were *in-situ* transferred to the XPS chamber. The XPS measurements were performed at room temperature under the UHV of $\sim 8 \times 10^{-8}$ Pa using an Al $K\alpha$ source ($h\nu = 1486.6$ eV) and a hemispherical electron energy analyzer, SCIENTA SES2002.

Figure 1 shows wide range XPS spectra of the samples. After the SR irradiation, the intensity of C 1s core level is stronger than that of Au 4f core level. After the initial UV irradiation in oxygen for ~ 4 hours [O_2+UV (~ 4 h)], the intensity of C 1s core level decreases while that of O 1s core level increases. It should be noted that neither UV irradiation in vacuum (without O_2) nor O_2 exposure changed XPS spectra significantly (not shown). Further UV irradiation in oxygen [O_2+UV (~ 90 h)] makes the intensities of Au core-levels prominent, indicating that the carbon materials on the optics can be removed. It

is noteworthy that the relative intensity of O 1s and Au 4p core-levels does not increase but decreases though the intensity of O 1s core level becomes stronger than that of C 1s core level.

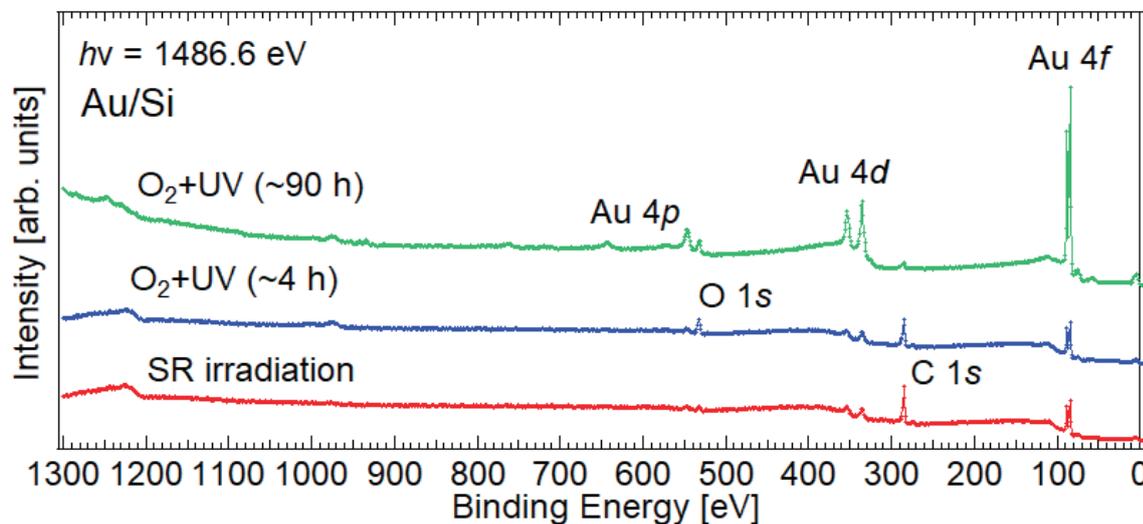


Fig. 1 Wide range XPS spectra of Au/Si surfaces.

Figure 2 shows normalized C 1s XPS spectra of the samples. The C 1s peak after the SR irradiation is located at ~ 284.5 eV and can be assigned to graphitic C=C and/or C-C bonds [2]. Since the so-called contamination C 1s peak-position was different (not shown), graphite-like carbon materials are found to be produced on the Au-coated Si samples after the SR irradiation, as previously reported [1]. After the initial UV irradiation in oxygen [O_2+UV (~ 4 h)], a new structure appears around the binding energy of ~ 289 eV, indicating that the C=O and/or O-C=O bonds [2] form during the UV/ozone cleaning. This spectral change confirms that the carbon on the surface does react with the oxygen radicals created by UV/ozone. These two components remain after further UV irradiation in oxygen [O_2+UV (~ 90 h)] although the intensity ratio of two components and the total intensity (Fig. 1) change.

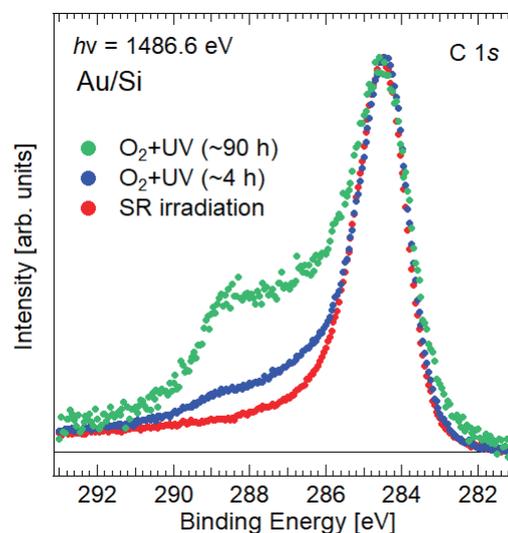


Fig. 2 C 1s XPS spectra of Au/Si surfaces.

References

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- [2] X. Chen, X. Wang, and D. Fang, *Fuller. Nanotub. Carbon Nanostructures*, **2020**, *28*, 1048.