

Dielectric relaxation of thin films of polyamide random copolymers

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(Received 13 February 2015; published xxxxxx)

We investigate the relaxation behavior of thin films of a polyamide random copolymer, PA66/6I, with various film thicknesses using dielectric relaxation spectroscopy. Two dielectric signals are observed at high temperatures, the α process and the relaxation process due to electrode polarization (the EP process). The relaxation time of the EP process has a Vogel-Fulcher-Tammann type of temperature dependence, and the glass transition temperature, T_g , evaluated from the EP process agrees very well with the T_g determined from the thermal measurements. The fragility index derived from the EP process increases with decreasing film thickness. The relaxation time and the dielectric relaxation strength of the EP process are described by a linear function of the film thickness d for large values of d , which can be regarded as experimental evidence for the validity of attributing the observed signal to the EP process. Furthermore, there is distinct deviation from this linear law for thicknesses smaller than a critical value. This deviation observed in thinner films is associated with an increase in the mobility and/or diffusion constant of the charge carriers responsible for the EP process. The α process is located in a higher-frequency region than the EP process at high temperatures but merges with the EP process at lower temperatures near the glass transition region. The thickness dependence of the relaxation time of the α process is different from that of the EP process. This suggests that there is decoupling between the segmental motion of the polymers and the translational motion of the charge carriers in confinement.

DOI: [10.1103/PhysRevE.00.002600](https://doi.org/10.1103/PhysRevE.00.002600)

PACS number(s): 82.35.-x, 77.22.Ch, 81.05.Lg

I. INTRODUCTION

Amorphous materials exhibit glass transition behavior when cooled from a high temperature to a low temperature under appropriate cooling conditions [1]. At the glass transition temperature, T_g , the motion of the α process is almost frozen, such that the characteristic time of the motion extends to a macroscopic time scale. For polymeric systems, the physical origin of the α process is attributed to the segmental motion of the polymer chains. The freezing of the α process can usually be explained by an anomalous increase in the scale of the characteristic length of the dynamics when approaching the glass transition temperature [2]. Dynamical heterogeneity is strongly correlated with this increase in the scale of the characteristic length and can be regarded as the most important concept in elucidating the mechanism of the glass transition [3]. Glass transitions in confined geometry, such as thin polymer films and small molecules in nanopores, have been widely investigated to determine the scale of the characteristic length in the glass transition dynamics [4]. Recent measurements show that for thin polymer films there is a large deviation in T_g and related dynamics from the bulk, albeit with some exceptions [5,6].

Being related to the heterogeneous dynamics near the glass transition temperature, the correlation between two different modes of molecular motion has been investigated [7,8]. For example, decoupling between the translational and rotational motions is usually observed in supercooled liquid states near T_g . Similar decoupling of the translational and rotational

motions has also been observed in confinement on the nanoscale [9].

The dependence of the glass transition on the film thickness has been investigated mainly for nonpolar polymers, such as polystyrene, poly(methyl methacrylate), and so on [10,11]. In contrast, the glass transition dynamics of thin films of polar polymeric systems have not yet been investigated intensively, although there have been some reports on the dependence of glassy dynamics on the film thickness for thin films of strong polar polymers such as polysulfone [12] and poly(bisphenol A carbonate) [13]. Polyamides are one of these strong polar polymers [14]. Because of their industrial importance, many experimental investigations, including dielectric measurements, have been carried out on polyamides over the past 50 yr [15–20]. The dielectric spectra of the polyamide family exhibit many processes, including γ , β , and α processes related to the molecular motion of polymer chains during heating from a low temperature to a high temperature of up to 140 °C. These processes are generally attributed to local motions of the methylene groups (γ process), rotations of the amide groups (β process), or segmental motions (α process) [14]. In addition to these three processes, other large dielectric signals are observed in high-temperature and low-frequency regions and should be related to the motion of the charge carriers within the samples. The dc conductivity process, the process related to the Maxwell-Wagner-Sillars (MWS) interfacial polarization, and the electrode polarization process (EP process) are usually observed [21]. For polyamides, the charge carrier related to the three processes is mainly charged hydrogen, i.e., protons, which come from the amide linkage formed via hydrogen bonding between the carbonyl group and the amino group [15]. For amorphous polymers, there is

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no interface between the crystalline and amorphous phases. Hence, MWS polarization process is not expected to be a major process in the low-frequency and high-temperature region for amorphous polymers. In this case, the EP process might be more important than MWS polarization process. The molecular origin of the electrode polarization is the partial blocking of charge carriers at the sample-electrode interface. This leads to a separation of the positive and negative charges, giving rise to additional polarization, as further explained in Sec. III.

Recent dielectric measurements of nylon 1010 [19] show that the dependence of the relaxation time of the EP process on temperature can be described by the Vogel-Fulcher-Tammann (VFT) law. This suggests the possibility of monitoring the glassy dynamics through investigation of the EP process. Hence, thin films of polyamide systems are expected to be suitable for the simultaneous investigation of glassy dynamics and the motion of charge carriers, as well as the correlation between the two processes. Because the translational motion of charge carriers can be determined from the EP process, polyamides can be regarded as suitable systems for investigating the correlation between the segmental motion of polyamide chains and the diffusion motion of charge carriers in the polyamide matrix.

In this study, we investigate the dielectric behavior of thin films of an aromatic polyamide random copolymer with various film thicknesses. Dielectric relaxation spectroscopy is used to elucidate the dependence of the α process and the EP process on the film thickness, which could lead to a better understanding of the relationship between the segmental motion and the motion of charge carriers in confinement. Following this Introduction, the experimental details are given in Sec. II, and the theoretical background, especially on the EP process, is given in Sec. III. In Sec. IV, the experimental results from dielectric relaxation spectroscopy are given, along with an analysis in terms of the model developed by Coelho. Finally, a brief summary is given in Sec. VI.

II. EXPERIMENTS

A. Samples

The aromatic polyamide random copolymer PA66/6I was supplied by Solvay [20]. The weight-averaged molecular weight of the copolymer is $M_w = 1.5 \times 10^4$, and the distribution of the molecular weight is given by $M_w/M_n = 1.6$. The chemical formula is schematically shown in Fig. 1. The glass transition temperature, T_g , of the original material was measured using differential scanning calorimetry (DSC), and was found to be 357 K under heating at a rate of 10 K/min.

The following procedures were carried out to prepare thin films for dielectric measurements. The polyamide was first

dissolved in 1,4-butanediol at 200 °C to obtain solutions of PA66/6I at concentrations of 2 wt.%, 1 wt.%, and 0.5 wt.% [22]. The solutions were then diluted at 160 °C with the same amount of chloroform. Thin films were subsequently prepared on an aluminum-deposited glass substrate from the diluted solution using spin coating. Following that, aluminum was again vacuum deposited to serve as an upper electrode. The sample was then annealed at 160 °C for 12 h before the dielectric measurements. The thickness of the film was controlled through the solution concentration and the speed of rotation during spin coating. The absolute film thickness was determined using atomic force microscopy measurements.

B. X-ray scattering measurements

X-ray scattering measurements were performed at SPring-8 BL40B2. A simultaneous measurement system utilizing small-angle x-ray scattering (SAXS) and wide-angle x-ray scattering (WAXS) was used. The measurement conditions were as follows: The camera lengths were 1757.4 and 90.0 mm for SAXS and WAXS, respectively; the x-ray wavelength was 0.9 Å; the detector system was a CCD camera with an image intensifier for SAXS, while that for WAXS was a flat panel.

C. Dielectric relaxation spectroscopy

Dielectric relaxation spectroscopy (DRS) measurements were made using an LCR meter (Agilent Technology, E4980A, 4292A) and Novocontrol AKB analyzer. The measured frequency range was from 0.01 Hz to 2 MHz and the temperature range was from 93 K to 423 K. The complex electric capacitance $C_{\text{mes}}^*(\omega)$ was obtained with the DRS measurements, where $\omega = 2\pi f$ and f is the frequency of the applied electric field. Because of the electric resistance R of the electrode formed by the vacuum deposition of aluminum, there is an additional contribution to the electric capacitance in the high-frequency region [23]. This contribution can be corrected using the assumption that the sample condenser can be described as a serial circuit of the condenser $C^*(\omega)$ and R [10]. Here the following relation holds:

$$C^*(\omega) = \frac{C_{\text{mes}}^* + i\omega R|C_{\text{mes}}^*|}{1 - 2\omega R\text{Im}(C_{\text{mes}}^*) + \omega^2 R^2|C_{\text{mes}}^*|^2}. \quad (1)$$

The complex electric capacitance $C^*(\omega)$ thus obtained is described by $C^* \equiv C_0 \varepsilon^*(\omega)$, where C_0 is the geometrical capacitance and $\varepsilon^*(\omega)$ is the complex dielectric permittivity of the sample. Here $C_0 = \varepsilon_0 \frac{S}{d}$, where ε_0 is the dielectric permittivity *in vacuo*, S is the area of the sample or electrode, and d is the thickness of the sample or the distance between the two electrodes. In the present measurements, $S = 8 \times 10^{-6} \text{ m}^2$. The complex dielectric permittivity is given by

$$\varepsilon^*(\omega) = \varepsilon'(\omega) - i\varepsilon''(\omega), \quad (2)$$

where ε' and ε'' are the real and imaginary parts of the complex dielectric permittivity, respectively. The voltage applied to the samples for dielectric measurements was selected as follows: 1.0 V for $d = 556 \text{ nm}$; 0.5 V for $d = 114, 99, 56$, and 40 nm ; and 0.2 V for $d = 20 \text{ nm}$. Hence, the applied electric field E ranges from 2 to 12 MV/m for the present measurements.

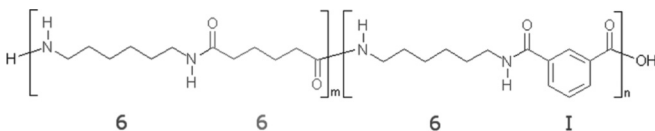


FIG. 1. The chemical formula of the amorphous polyamide copolymer PA66/6I.

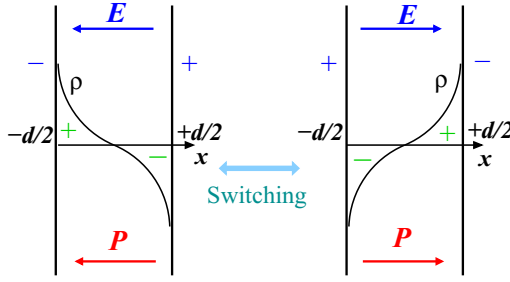


FIG. 2. (Color online) Distribution of the charge between two electrodes under an applied electric field, with switching of the macroscopic dipole moment by reversal of the electric field. This figure has been modified from the original by Coelho.

III. THEORETICAL MODELS OF THE ELECTRODE POLARIZATION PROCESS

There are several theoretical models of the EP process [24–26]. According to these theoretical models, the relaxation time of the electrode polarization, τ_{ep} , and the relaxation strength of the electrode polarization, $\Delta\epsilon_{ep}$, can be described by a linear function of the distance between the two electrodes, positive and negative, which corresponds to the thickness of the thin film, d .

If there are mobile charge carriers in a sample, then the spatial distribution of the charges changes depending on changes in the applied electric field. In the framework of the Debye theory, the dipolar relaxation can be described as a process associated with viscous forces originating from the matrix, where the charge carriers move randomly with thermal fluctuations. In the absence of an applied electric field, the charge carriers are uniformly distributed in the sample, and the sample is thus almost electrically neutral. In the presence of a dc electric field, however, the charge carriers are separated depending on their polarity, and a heterogeneous charge distribution appears within the sample. If the direction of the applied electric field is reversed, then the charge carriers move in such a way that a new heterogeneous charge distribution is reached, as shown in Fig. 2. For an ac electric field, the dynamics of the charge carriers have been modeled by Coelho [25], and the complex dielectric permittivity has been derived for applied electric fields with angular frequency ω as

$$\epsilon^*(\omega) = \epsilon_\infty \frac{1 + i\omega\tau}{i\omega\tau + \frac{\tanh Y}{Y}}, \quad (3)$$

where

$$Y \equiv \frac{d}{2L} \sqrt{1 + i\omega\tau}, \quad (4)$$

$$L \equiv \sqrt{D\tau}. \quad (5)$$

Here τ is the relaxation time characterizing the relaxation phenomena of the charge distribution or the electrode polarization upon changes in the external applied electric field, D is the diffusion constant of the charge carriers, and L is the Debye length, i.e., the scale of the characteristic length of the electrostatic double layer near the electrodes. In addition,

relations among τ , μ , σ , and n_0 are obtained,

$$\tau^{-1} = \frac{\mu en_0}{\epsilon_\infty \epsilon_0} = \frac{\sigma}{\epsilon_\infty \epsilon_0}, \quad (6)$$

$$\sigma \equiv \mu en_0, \quad (7)$$

where σ is the dc conductivity, n_0 is the equilibrium concentration of charge carriers, e is the elementary electric charge, μ is the mobility of the charge carrier, and ϵ_∞ is the dielectric permittivity at very high frequency. Applying the Einstein relation leads to a relation between D and μ ,

$$D = \frac{\mu k_B T}{e}, \quad (8)$$

where k_B is the Boltzmann constant, and T is the absolute temperature [27]. This equation is an expression of the fluctuation-dissipation theorem [28]. Hence, the Debye length can be expressed as

$$L = \frac{1}{e} \left(\frac{\epsilon_\infty \epsilon_0 k_B T}{n_0} \right)^{1/2}. \quad (9)$$

If $\omega\tau \approx 0$ and $\delta \equiv d/2L \gg 1$, then the complex dielectric permittivity can be described by

$$\epsilon^*(\omega) \approx \epsilon_\infty \left(1 + \frac{\delta}{1 + i\omega\tau\delta} \right). \quad (10)$$

In real polymeric materials, there is a distribution of the relaxation times because of the heterogeneity of the materials. The dielectric permittivity due to the EP process can be expressed as

$$\epsilon^*(\omega) = \epsilon_\infty + \epsilon_{ep}^*(\omega), \quad (11)$$

$$\epsilon_{ep}^*(\omega) = \frac{\Delta\epsilon_{ep}}{1 + (i\omega\tau_{ep})^{\alpha_{ep}}}, \quad (12)$$

where

$$\Delta\epsilon_{ep} = \epsilon_\infty \delta = \epsilon_\infty \frac{d}{2L}, \quad (13)$$

$$\tau_{ep} = \tau\delta = \frac{\epsilon_\infty \epsilon_0}{\mu en_0} \frac{d}{2L}. \quad (14)$$

Using Eqs. (9), (13), and (14), we can obtain equations for the mobility $\mu(T)$ and the equilibrium concentration of charge carriers $n_0(T)$,

$$n_0(T) = \frac{4\epsilon_0 k_B T}{e^2 \epsilon_\infty d^2} (\Delta\epsilon_{ep})^2, \quad (15)$$

$$\mu(T) = \frac{e \epsilon_\infty d^2}{4k_B T} \frac{1}{\Delta\epsilon_{ep} \tau_{ep}}, \quad (16)$$

which relate $\mu(T)$ and $n_0(T)$ to the experimentally observed values of $\Delta\epsilon_{ep}(T)$ and $\tau_{ep}(T)$. Here τ_{ep} is the relaxation time of the EP process. Furthermore, the diffusion constant of the charge carriers, $D(T)$, can be determined using Eqs. (8), (13), and (14) from the values of $\Delta\epsilon_{ep}$ and τ_{ep} as follows:

$$D(T) = \frac{\epsilon_\infty d^2}{4\Delta\epsilon_{ep} \tau_{ep}}. \quad (17)$$

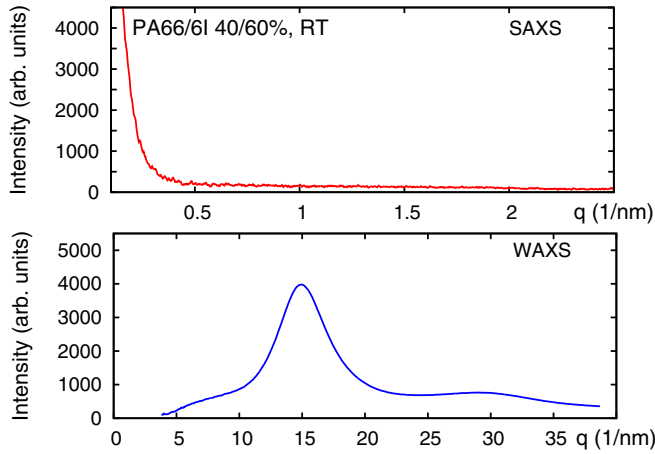


FIG. 3. (Color online) X-ray scattering intensity with q ranging from 0.1 to 38 nm^{-1} at room temperature for the amorphous polyamide copolymer PA66/6I. The top figure is for the SAXS region, and the bottom one is for the WAXS region.

IV. RESULTS AND DISCUSSION

A. X-ray scattering measurements

Figure 3 shows the dependence of the x-ray scattering intensity on the scattering vector q at room temperature for the amorphous polyamide copolymer PA66/6I. The scattering profiles for both the SAXS and the WAXS regions are shown in Fig. 3. In the WAXS region, for $q > 5 \text{ nm}^{-1}$, there is a broad peak that is usually called an amorphous halo. This peak originates from the short-range order of the structure. The amorphous halo is one typical scattering pattern for amorphous materials. In the SAXS region, there is no peak, only a continuous decay from $q = 0$. Hence, we can infer that there is neither a higher-ordered structure, such as a lamellar structure, nor a crystalline structure in PA66/6I.

B. Dielectric spectra in the temperature domain

Figure 4 shows the dependence of the complex dielectric permittivity on the temperature, measured at various frequencies for the amorphous polyamide copolymer PA66/6I at a 680-nm thickness. In Fig. 4(a), there are several contributions to the dielectric permittivity depending on the temperature and frequency. There is a dielectric loss peak due to the γ process located at around 170 K for 20 Hz, and there is also a broad contribution due to the β process between 220 K and 280 K. In the low-frequency and high-temperature region, there is a very large contribution, which should be related to the motion of charge carriers. Here this behavior can be attributed to the EP process and/or conductivity, as there should not be a crystal-amorphous interface in this amorphous polyamide copolymer. A detailed discussion on attributing this to the EP process is given in Sec. IV D. In addition to the three contributions, there is the α process as a shoulder of the large peak of the EP process. In Fig. 4(b), we can also see that there are signals in the real part of the dielectric permittivity, which correspond to the four different dynamical contributions. It should be noted that the observed data in Fig. 4 were obtained after heating above 100 °C. In other words, the results in Fig. 4

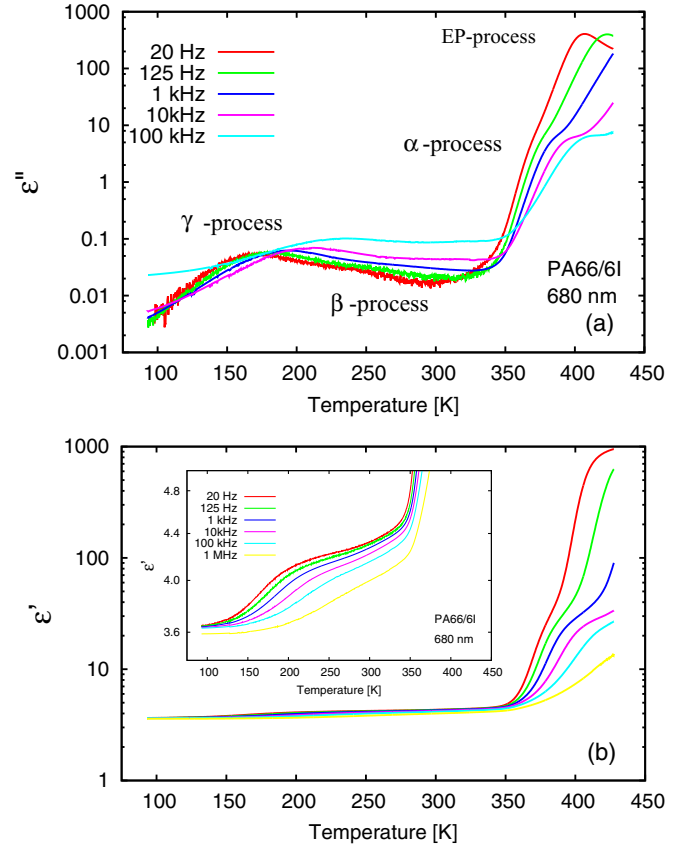


FIG. 4. (Color online) The dependence of the complex dielectric permittivity on the temperature at various frequencies, for films of the amorphous polyamide copolymer PA66/6I with a thickness of 680 nm. (a) The dielectric loss ϵ'' ; (b) the real part ϵ' ; and (inset) a magnified image of ϵ' .

should correspond to those for the dry state [17,20]. In this paper, we concentrate on the α process and the EP process.

C. Dielectric spectra in the frequency domain

For strong polar materials, the EP process may show a strong nonlinear effect against the applied electric field. Figure 5 shows the real and imaginary parts of dielectric permittivity in the frequency domain for various applied voltages from 0.1 to 0.5 V in thin films of PA66/6I with a 40-nm thickness. In Fig. 5, the value of E ranges from 2.5 to 12.5 MV/m, in which the selected range of E in the present measurements is almost included. The curves observed for various applied voltages are overlapped with each other, as shown in Fig. 5. From these measurements, we can estimate possible maximum deviation due to the change in electric field as follows: For the peak frequency of the EP process f_{ep} at a given temperature, possible deviation $\Delta f_{ep}/f_{ep}$ is less than 0.02, and for the peak value of ϵ'' of the EP process, deviation is less than 3%. Therefore, the observed change in peak position and height of the EP process with change in the film thickness is much larger than these possible deviations. Therefore, it can be justified that experimental results in this paper do not depend on the value of E for the range of E selected in the present measurements.

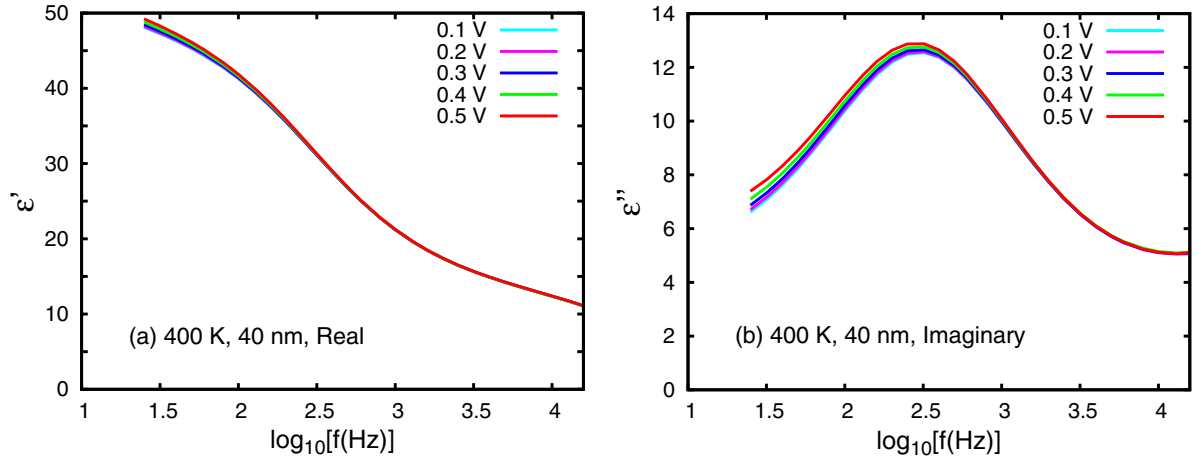


FIG. 5. (Color online) The dependence of the real and imaginary parts of the complex dielectric permittivity on the frequency at 400 K for various applied voltages from 0.1 to 0.5 V in thin films of the amorphous polyamide copolymer PA66/6I with a 40-nm thickness.

Figure 6 shows the dependence of the complex dielectric permittivity on the frequency for thin films of the amorphous polyamide copolymer PA66/6I with a 556-nm thickness. In Fig. 6, different processes including the α process and the EP process contribute to the dielectric permittivity. These components are readily observed over the temperature range

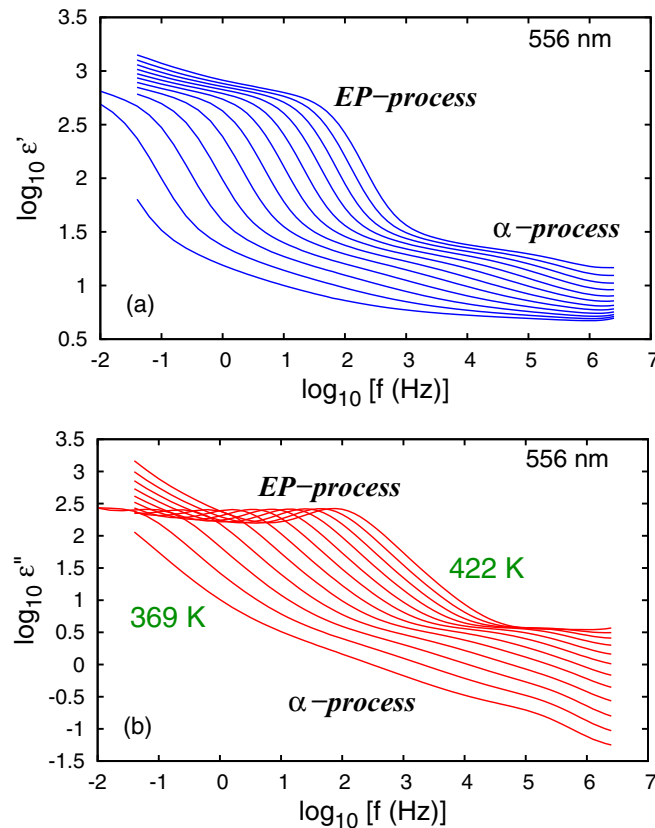


FIG. 6. (Color online) The dependence of the complex dielectric permittivity on the frequency for thin films of the amorphous polyamide copolymer PA66/6I with a thickness of 556 nm. (a) The real part of ϵ^* and (b) the imaginary part of ϵ^* . The temperature ranges from 422.8 K (right) to 369.5 K (left).

from 369 K to 422 K. Furthermore, we also show the dependence of the complex conductivity $\sigma^*(=\sigma' + i\sigma'')$ on the frequency for the thin films. Here σ' and σ'' are the real and imaginary parts of the complex conductivity, respectively, and the values of σ' and σ'' are evaluated from the following relations: $\sigma' = \epsilon_0 \epsilon'' \omega$ and $\sigma'' = \epsilon_0 \epsilon' \omega$. Figure 7 shows the dependence of the logarithm of the real and imaginary parts of the complex conductivity observed at 422 K for thin films of PA66/6I with a 556-nm thickness. The value of σ' approaches a constant value (plateau value) at ca. 10^4 Hz with decreasing frequency, although there is no exact plateau but with a small slope because of the overlap of the α process. On the other hand, the value of σ'' shows the signal of the onset of EP process at 10^3 Hz. According to the usual interpretation on the EP process in the conductivity representation proposed by Kremer and co-workers [29–31], it is reasonable to regard the “plateau” of σ' at 10^4 Hz as the onset of dc conductivity, which should be required for the appearance of the EP process. If this interpretation is valid, the physical origin of the increase

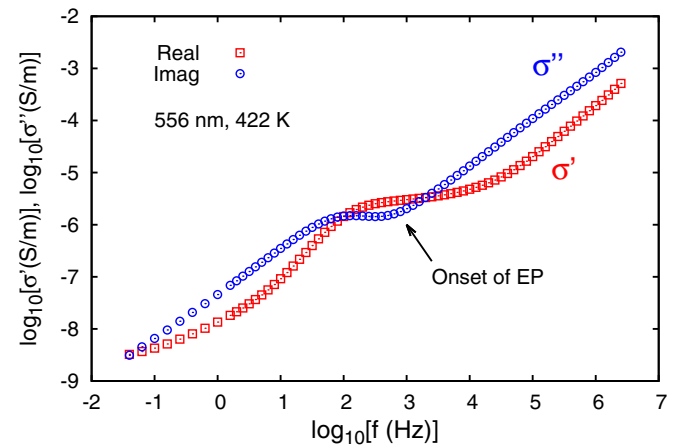


FIG. 7. (Color online) The dependence of the logarithm of the real and imaginary part of complex conductivity observed at 422 K for thin films of PA66/6I with a 556-nm thickness. The arrow shows the onset of the EP process.

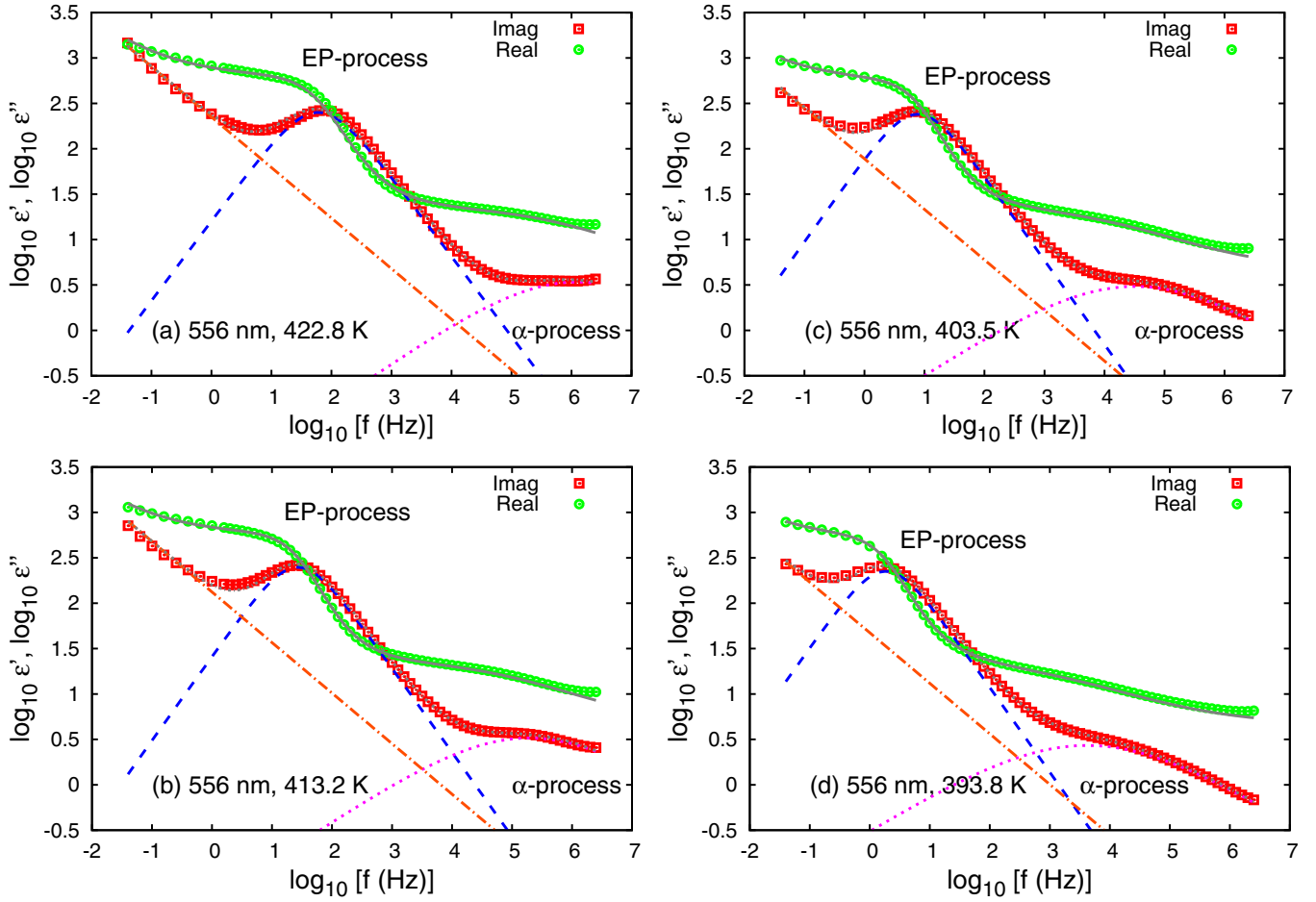


FIG. 8. (Color online) The dependence of the real and imaginary parts of the complex dielectric permittivity on the frequency at various temperatures—(a) 422.8 K, (b) 413.2 K, (c) 403.5 K, and (d) 393.8 K—for thin films of the amorphous polyamide copolymer PA66/6I with a 556-nm thickness showing (red squares) the imaginary part ϵ'' and (green circles) the real part ϵ' . The three different contributions to the imaginary part are also shown: The orange dashed-dotted line corresponds to the low-frequency component ϵ_{σ}^* , the blue dashed line corresponds to the EP process, and the pink dotted line corresponds to the α process. The curves for the three components and the two observed values of ϵ' and ϵ'' were evaluated by fitting the observed values to Eq. (18).

in ϵ' and ϵ'' in the low-frequency region should not be due to usual conductivity. Wang *et al.* reported that there is an increase in ϵ' and ϵ'' in the region lower in frequency than the location of the EP process [32]. Furthermore, they attributed this contribution to imperfectness of the blocking at electrodes of the charge carriers which are responsible for the EP process. This effect causes a deviation from the expected values from the theoretical model of the EP process. Here we adopt this interpretation for the increase in ϵ' and ϵ'' with decreasing frequency in low-frequency region. In this case, the contributions in the low-frequency region in ϵ' and ϵ'' are not directly related to the usual dc conductivity.

The frequencies at which the dielectric loss peaks due to the EP and α processes are located are shifted from the lower frequency region to the higher frequency region with increasing temperature, as shown in Fig. 6. In order to investigate the dielectric properties of the components separately, the observed dependence of ϵ' and ϵ'' on the frequency was reproduced by the model function

$$\epsilon^*(\omega) = \epsilon_{\infty} + \epsilon_{\sigma}^*(\omega) + \epsilon_{\alpha}^*(\omega) + \epsilon_{\text{ep}}^*(\omega), \quad (18)$$

where $\epsilon_{\text{ep}}^*(\omega)$ is the complex dielectric permittivity due to the EP process described by Eq. (12), $\epsilon_{\alpha}^*(\omega)$ is the dielectric permittivity due to the α process, and $\epsilon_{\sigma}^*(\omega)$ is the dielectric permittivity due to the imperfectness of the blocking at electrodes of the charge carriers. For the component of ϵ_{σ}^* , we use simple power-law functions for both real and imaginary parts of ϵ_{σ}^* ,

$$\text{Re}[\epsilon_{\sigma}^*](\omega) = A\omega^{-\tilde{m}}, \quad \text{Im}[\epsilon_{\sigma}^*](\omega) = B\omega^{-\tilde{m}'}, \quad (19)$$

where A , B , \tilde{m} , and \tilde{m}' are constants. These functions work well, as shown in Fig. 8. As for the α process, we adopt the Havriliak-Negami equation [33],

$$\epsilon_{\alpha}^*(\omega) = \frac{\Delta\epsilon_{\alpha}}{[1 + (i\omega\tau_{\alpha})^{\alpha_{\alpha}}]^{\beta_{\alpha}}}, \quad (20)$$

where $\Delta\epsilon_{\alpha}$ is the dielectric relaxation strength, α_{α} and β_{α} are shape parameters, and τ_{α} is the relaxation time of the α process. In the present analysis, the parameter β_{α} is fixed at unity because of a reduction in the free fitting parameters.

Figure 8 shows the dependence of ϵ' and ϵ'' on the frequency at four different temperatures—422.8 K, 413.2 K, 403.5 K,

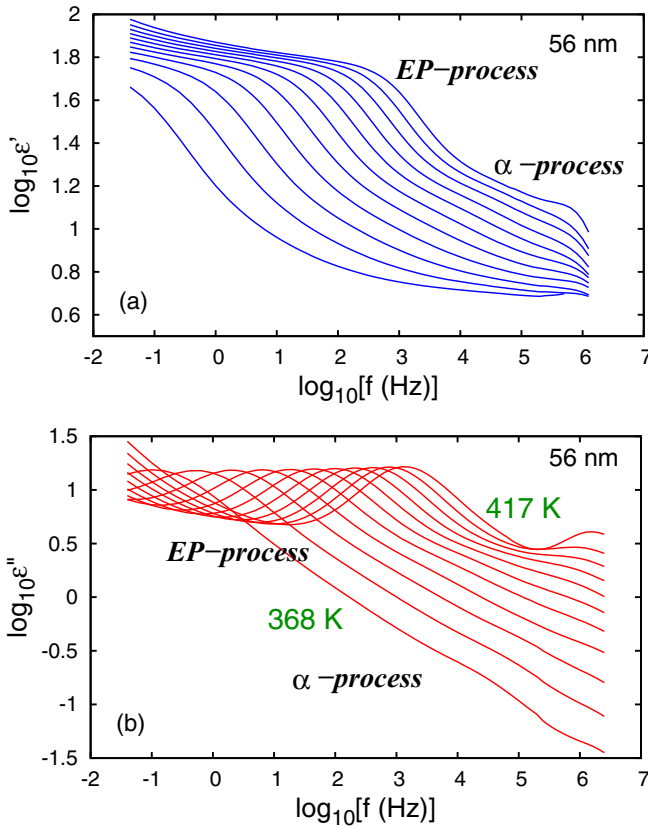


FIG. 9. (Color online) The dependence of the complex dielectric permittivity on the frequency for thin films of the amorphous polyamide copolymer PA66/6I with a thickness of 56 nm. (a) The real part of ϵ^* and (b) the imaginary part of ϵ^* . The temperature ranges from 417.6 K (right) to 368.7 K (left).

and 393.8 K—for a 556-nm-thick thin film of the amorphous polyamide copolymer PA66/6I. Calculated curves for the different components are also shown, and were obtained by fitting the observed data to the model function, Eq. (18). From Fig. 8, we can see that the observed dielectric permittivity is very well reproduced by the present model function and, as a result, several important physical quantities can be evaluated as a function of the temperature and film thickness.

Figure 9 shows the dependence of the complex dielectric permittivity on the frequency for a 56-nm-thick thin film of the amorphous polyamide copolymer PA66/6I. In Fig. 9, we can see that the peak height of the EP process in the curve of ϵ'' vs f is much smaller for the 56-nm-thick film than for the 556-nm-thick film. Furthermore, the peak positions of the EP process in the frequency domain at a given temperature are clearly shifted to the higher frequency region for the 56-nm-thick thin film, as compared to the 556-nm-thick film. In addition, for the 56-nm-thick film, the present model function again accurately reproduces the observed dielectric permittivity as a function of frequency for various temperatures, as shown in Fig. 10.

D. The electrode polarization process

Figure 11 shows the dependence of the relaxation rate of the EP process, f_{ep} , on the temperature, for thin films of the amorphous polyamide copolymer PA66/6I of various

thicknesses, ranging from 20 to 556 nm. The value of f_{ep} is evaluated from the relation $f_{ep} = (2\pi\tau_{ep})^{-1}$ with the best-fitted values of τ_{ep} obtained using Eq. (12). Here, the parameter β_{ep} is fixed to 1, such that the resulting frequency f_{ep} is equal to the frequency at which ϵ'' exhibits a peak. In Fig. 11, we can see that the relaxation rate of the EP process has a stronger temperature dependence than the Arrhenius type of temperature dependence. As shown by the curves in Fig. 11, the temperature dependence of f_{ep} is well reproduced by the VFT law [34–37],

$$\tau_{ep}(T) = \tau_{ep,0} \exp\left(\frac{U}{T - T_0}\right), \quad (21)$$

where U is a positive constant and T_0 is the Vogel temperature. At a given temperature, the relaxation rate of the EP process increases with decreasing film thickness. In other words, the relaxation time decreases. The VFT law is usually valid for the temperature dependence of the relaxation time of the α process or the normal mode [38], and the slowing down of the dynamics when approaching the Vogel temperature is essential to glassy dynamics. Therefore, we can expect that the molecular motion of the EP process is strongly associated with the molecular motion, the segmental motion, or the normal mode of polyamide copolymer systems. This result is consistent with the results reported in Ref. [19].

In Fig. 11, there seems to be the crossing of the temperature dependence of the rate of the EP process f_{ep} and that of the α process f_{α} , especially for smaller values of d . In the region where the two relaxation rates f_{ep} and f_{α} come across, it is very difficult and almost impossible to separate the component of the α process from that of the EP process. The extrapolated curve of the α process using the VFT law intersects with the observed curve of the EP process in the low-frequency region. Because the α process induces the EP process, it is reasonable to imagine that the two curves will merge without intersection.

Figure 12 shows the dependence of the Vogel temperature T_0 on the film thickness, evaluated by fitting the observed temperature dependence of the relaxation rate of the EP process to the VFT law. Here the glass transition temperature T_g is also evaluated, such that the relaxation time of the EP process, τ_{ep} , is equal to 10^3 s at T_g , and the resulting T_g is plotted in Fig. 12. Furthermore, the glass transition temperature determined by DSC measurements is also plotted, for comparison with other T_g results. Here, for the bulk system, the T_g from DSC is 357 K, the T_g evaluated from the relaxation rate of the EP process is 357.3 ± 0.8 K, and the Vogel temperature T_0 is 299 ± 6 K. The T_g evaluated from the EP process agrees very well with that from the DSC measurements. The difference between T_0 and T_g is approximately -58 K, which is consistent with the empirical relation that T_0 is approximately 50 K lower than T_g [39]. Therefore, this result suggests that the relaxation rate of the EP process can be utilized to determine the dynamics of the segmental motion, that is, the glassy dynamics, especially near the glass transition region. In Fig. 12, we can see that both T_g and T_0 decrease slightly with decreasing film thickness.

As shown in the above, the temperature dependence of the relaxation time of the EP process is used to determine the glass transition temperature T_g of the amorphous

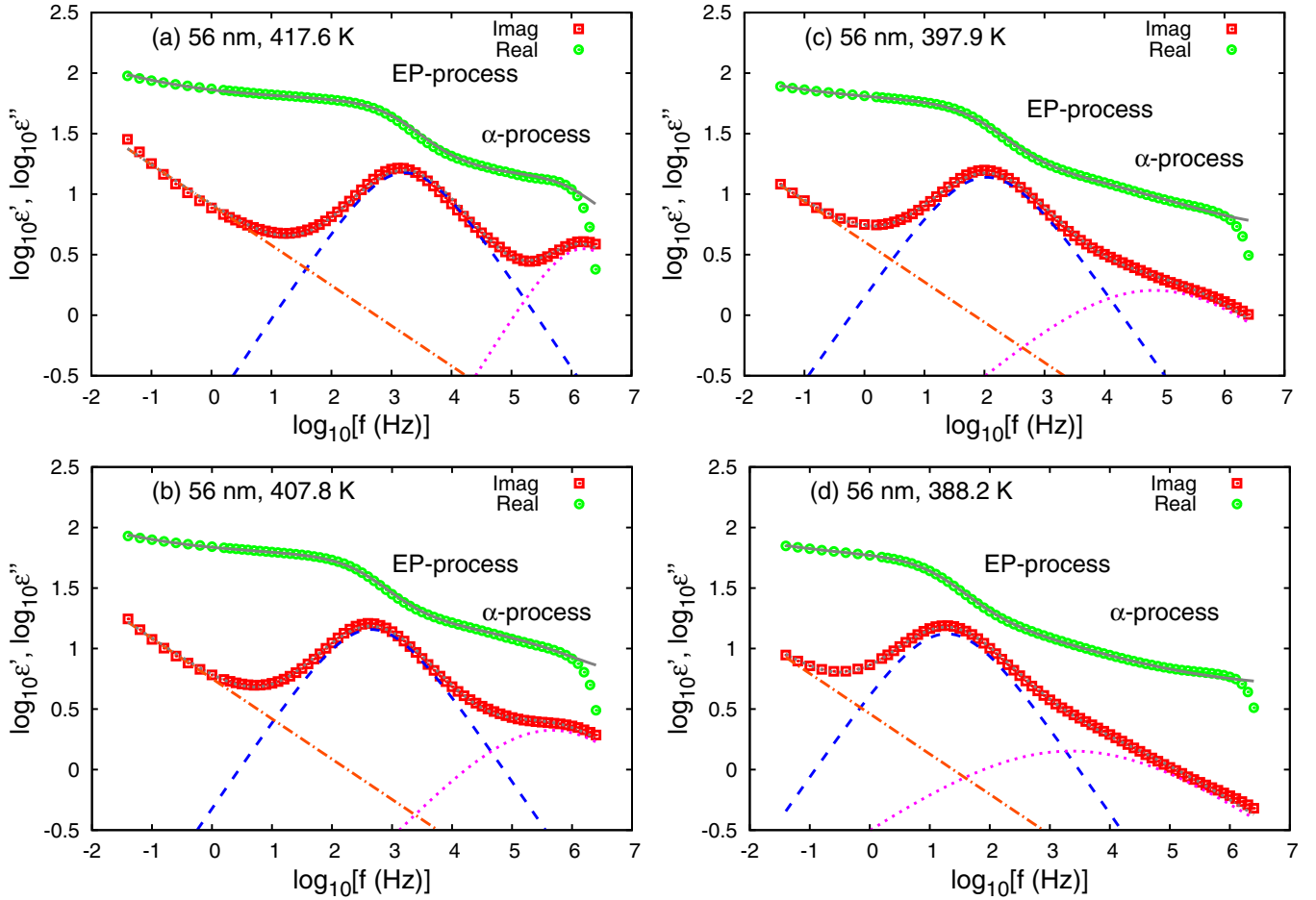


FIG. 10. (Color online) The dependence of the real and imaginary parts of the complex dielectric permittivity on the frequency at various temperatures—(a) 417.6 K, (b) 407.8 K, (c) 397.9 K, and (d) 388.2 K—for thin films of the amorphous polyamide copolymer PA66/6I with a 56-nm thickness. The meanings of the symbols and curves are identical to those in Fig. 8.

polyamide copolymer PA66/6I. Furthermore, we can use the observed temperature dependence of τ_{ep} to determine the fragility index, which characterizes the glassy dynamics of amorphous materials. Here we define the fragility index m as [40]

$$m = \left[\frac{d \log_{10} \tau_{ep}(T)}{d(T_g/T)} \right]_{T=T_g}, \quad (22)$$

where the relaxation time τ_{ep} is used instead of τ_{α} , which should be used in the formal definition of the fragility index.

Figure 13 shows the dependence of the apparent fragility index on the film thickness, evaluated using the temperature dependence of the relaxation time of the EP process for thin films of the amorphous polyamide copolymer PA66/6I. In Fig. 13, the fragility index increases with decreasing film thickness. Therefore, the glassy dynamics of these thin films are expected to become more fragile with decreasing film thickness, provided that this apparent fragility index can be regarded as the fragility index evaluated from the relaxation time of the α process. A more in-depth discussion on this issue is given in Sec. V.

The dependence of the dielectric relaxation strength on the film thickness at 403 K for the EP process is shown in Fig. 14 for thin films of the amorphous polyamide copolymer

PA66/6I. The dielectric relaxation strength of the EP process, $\Delta\epsilon_{ep}$, monotonically increases with increasing film thickness. Above $d = 100$ nm, the thickness dependence of $\Delta\epsilon_{ep}$ is well fitted by a straight line with a slope of unity, as shown by the red line in Fig. 14. According to the theoretical model of the electrode polarization, the dielectric relaxation strength of the EP process can be expressed as a linear function of the thickness of the sample, i.e., the distance between the two electrodes, as shown in Eq. (13). Therefore, the linear relationship between $\Delta\epsilon_{ep}$ and the film thickness can be regarded as strong evidence for the validity of attributing the EP process to the strongest dielectric loss signal observed in the present measurements. Furthermore, it should be noted that there is a distinct deviation of the thickness dependence of $\Delta\epsilon_{ep}$ from the straight line with a slope of unity. If $\Delta\epsilon_{ep}$ (τ_{ep}) can be expressed as a linear function of d , the Debye length L and the characteristic relaxation time of the electrode polarization τ should be independent of the film thickness. However, if this is not the case, the values of L and τ should depend on the film thickness. Therefore, this deviation implies that there is an intrinsic dependence on the film thickness for the physical mechanism of the EP process, for thin films with a thickness less than a critical thickness d_c ($60 \text{ nm} < d_c < 100 \text{ nm}$).

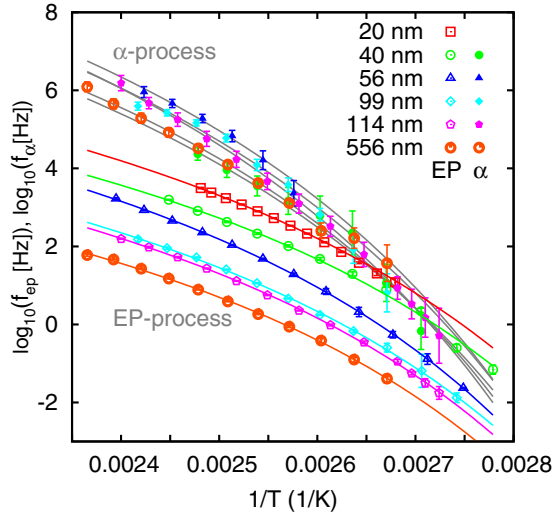


FIG. 11. (Color online) Dispersion map of the relaxation rate f_{ep} of the EP process and the relaxation rate f_{α} of the α process, for thin films of the amorphous polyamide copolymer PA66/6I with various thicknesses ranging from 20 to 556 nm. The values of f_{ep} and f_{α} are evaluated as $f_{ep} = (2\pi\tau_{ep})^{-1}$ and $f_{\alpha} = (2\pi\tau_{\alpha})^{-1}$, where τ_{ep} and τ_{α} are the fitting parameters obtained by fitting the observed ϵ^* to Eq. (18). The curves are governed by the VFT law. The curves of the α process were obtained under the condition that the Vogel temperature T_0 was fixed at the value obtained by fitting the temperature dependence of τ_{ep}^{-1} to Eq. (21).

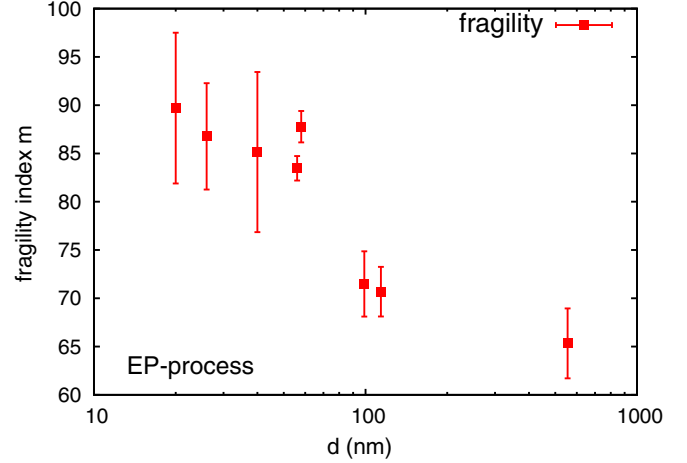


FIG. 13. (Color online) The dependence of the apparent fragility index m on the film thickness, based on the temperature dependence of the relaxation time of the electrode polarization process. Here T_g is defined as the temperature at which the relaxation time τ_{ep} is equal to 10^3 s.

relation:

$$f_{ep} \sim d^{-1}. \quad (23)$$

This result is consistent with Eq. (14) for the present model of the EP process. Below d_c , there is an intrinsic deviation from Eq. (14) in the same manner as observed for the dielectric relaxation strength of the EP process.

E. The α process

In this section, we discuss the experimental results on the relaxation time of the α process for thin films of the amorphous polyamide copolymer PA66/6I. The dielectric relaxation strength of the α process is much smaller than that of the EP process, as shown in Fig. 6. It is thus difficult to evaluate the physical parameters of the α process by extracting the

Figure 15(a) shows the dependence of the relaxation rate of the EP process, f_{ep} , on the film thickness—at 413 K, 393 K, and 373 K—for thin films of the amorphous polyamide copolymer PA66/6I. For a thickness larger than d_c , the dependence of the relaxation rate of the EP process on the film thickness is well described by the following

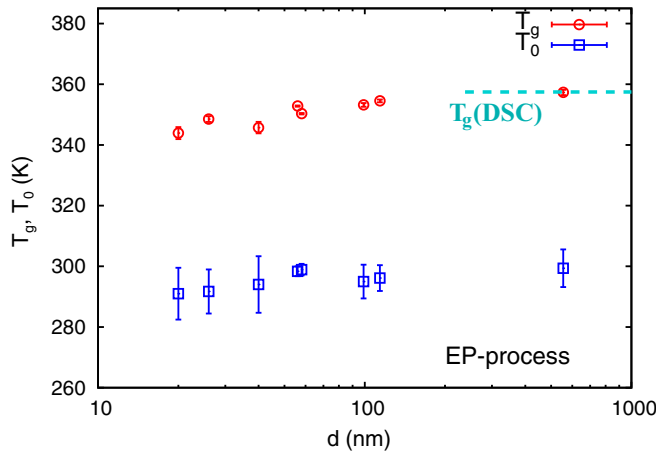


FIG. 12. (Color online) The dependence of the Vogel temperature T_0 (\square) and the glass transition temperature T_g (\circ) on the film thickness, as determined by the temperature dependence of the relaxation time τ_{ep} of the EP process for thin films of the amorphous polyamide copolymer PA66/6I. Here, T_g is defined such that $\tau_{ep}(T_g) = 10^3$ s. The value of T_g as evaluated by DSC measurements is also shown.

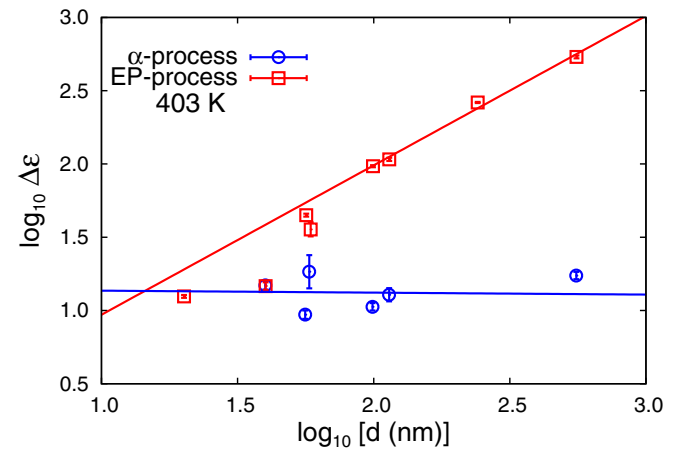


FIG. 14. (Color online) The dependence of the dielectric relaxation strength $\Delta\epsilon$ at 403 K on the film thickness for the EP process (\circ) and the α process (\square) in thin films of the amorphous polyamide copolymer PA66/6I with various thicknesses. The slope of the straight line for the EP process is equal to unity and that for the α process is almost equal to zero.

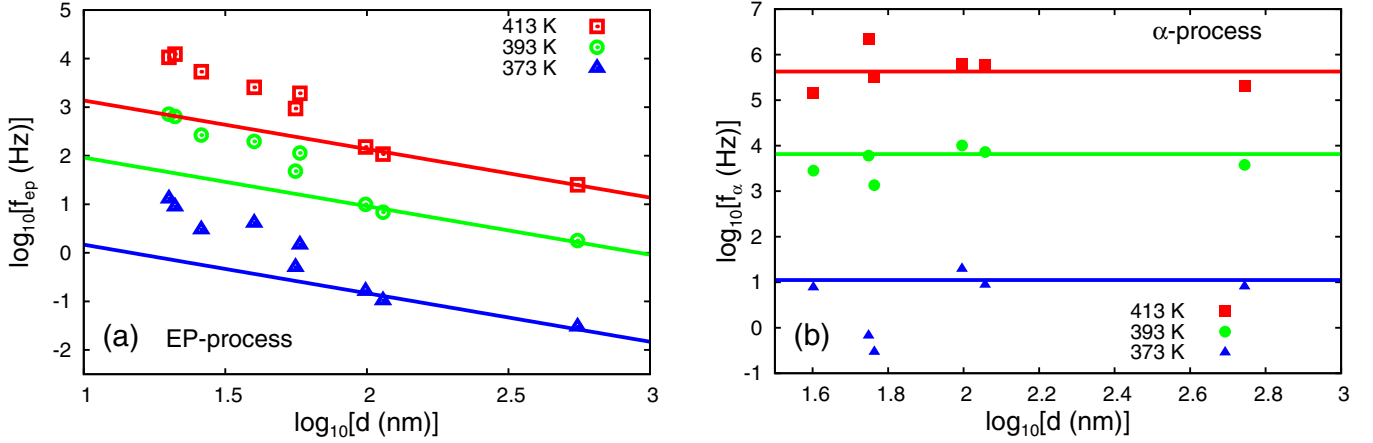


FIG. 15. (Color online) The dependence of (a) the relaxation rate f_{ep} of the EP process and (b) the relaxation rate f_{α} of the α process on the film thickness—at 413 K, 393 K, and 373 K—for thin films of PA66/6I with various thicknesses ranging from 556 to 20 nm. The straight lines in (a) were obtained by fitting for three values of the thin films of PA66/6I with a thickness larger than 99 nm. The slopes are fixed to be -1 .

component caused by the α process from the overall dielectric spectra. However, data fitting by Eq. (18) enables the extraction of only the component from the α process, as well as the elucidation of the temperature dependence of the relaxation rate of the α process, as shown in Fig. 11.

For thin films of PA66/6I with a given film thickness, the relaxation rate of the α process is located at a frequency region higher than that of the EP process at a given temperature. The deviation of the relaxation rate of the α process from that of the EP process becomes smaller as the temperature approaches the glass transition temperature. Hence, the α process appears to merge with the EP process with decreasing temperature. This is the reason for the difficulty in evaluating the relaxation rate of the α process. At the same time, the merging of the α process with the EP process can be regarded as an evidence that the EP process is strongly associated with the α process, especially at the glass transition region. In Fig. 11, the temperature dependence of the relaxation rate of the α process is well reproduced by the VFT law, under the condition that the Vogel temperature of the α process is the same as that of the EP process for each film thickness.

As for the dependence of the α process on the film thickness, the dielectric relaxation strength and the relaxation rate of the α process are shown as functions of the film thickness at a given temperature in Figs. 14 and 15(b), respectively. Although there is some scatter in the data points, they suggest that there is almost no systematic dependence of the dielectric relaxation strength and the relaxation rate of the α process on the film thickness, for the thickness range investigated in this study. The dependence of the relaxation rate of the α process on the film thickness in comparison to that of the EP process is discussed in Sec. V, for thin films of PA66/6I.

F. Relaxation of the distribution of space charge

As shown in Sec. IV D, the dielectric relaxation phenomenon of the EP process exhibits an interesting dependence on the temperature and film thickness, for the amorphous polyamide copolymer PA66/6I. Here we try to evaluate the in-

trinsic nature of the EP process, based on the theoretical model given in Sec. III. From the observed values of $\tau_{ep}(T)$ and $\Delta\epsilon_{ep}$, the temperature dependence of the mobility and equilibrium concentration of charge carriers within the polymeric system can be evaluated using Eqs. (15) and (16). Figure 16 shows this dependence for thin films of PA66/6I of various thicknesses. In Fig. 16, the equilibrium concentration of charge carriers n_0 shows a weak temperature dependence (figure inset), while the mobility $\mu(T)$ shows a strong temperature dependence that can be expressed by the VFT law of mobility as follows:

$$\mu = \mu_0 \exp\left(-\frac{\tilde{U}}{T - T_0}\right). \quad (24)$$

Here μ_0 and \tilde{U} are positive constants. The Vogel temperatures T_0 obtained from $\mu(T)$ are 294 K and 298 K for $d = 40$ nm

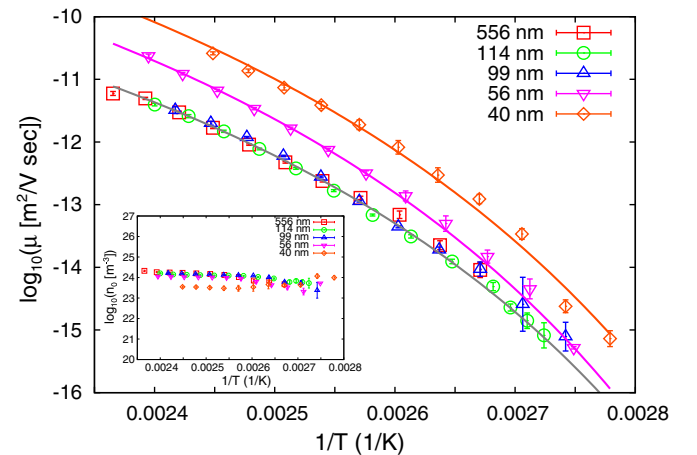


FIG. 16. (Color online) The dependence of the charge carrier mobility μ on the inverse of the absolute temperature $1/T$ for thin films of the amorphous polyamide copolymer PA66/6I with various thicknesses ranging from 556 to 40 nm. The curves were obtained by fitting the observed values to the VFT equation for the mobility, Eq. (15). In the inset, the dependence of the equilibrium concentration of charge carriers n_0 on $1/T$ is also given for the same thin films of the amorphous polyamide copolymer PA66/6I.

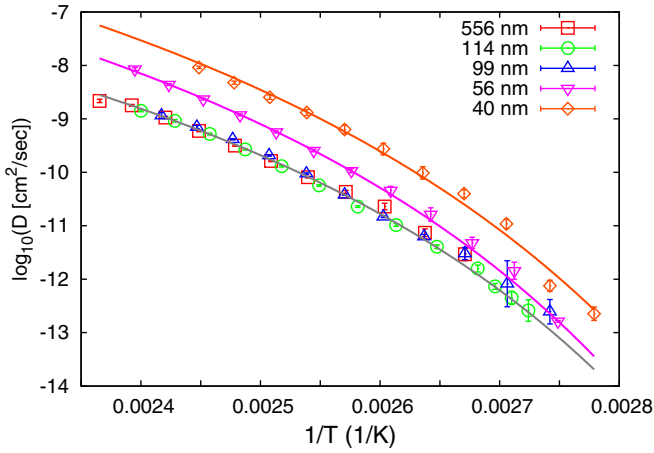


FIG. 17. (Color online) The dependence of the diffusion constant of the charge carriers D on the inverse of the absolute temperature $1/T$ for thin films of the amorphous polyamide copolymer PA66/6I with various thicknesses ranging from 556 to 40 nm.

and $d > 99$ nm, respectively. These values are nearly identical to those evaluated from $\tau_{ep}(T)$, as shown in Fig. 12.

For the thin films with thicknesses of 556, 114, and 99 nm, the temperature dependencies of the mobility agree well with each other and are well reproduced by the VFT law with the same parameters. However, for the thin films with thicknesses less than d_c , the mobility of the charge carriers at a given temperature increases with decreasing film thickness. This is consistent with the results of the relaxation rate of the EP process in Fig. 15, that is, the crossover of f_{ep} from the straight line with a slope of -1 to the line with a larger slope as the film thickness decreases. Therefore, we can conclude that *the intrinsic mobility of the charge carriers remains almost unchanged for a film thickness above a critical thickness d_c , while the mobility increases with decreasing film thickness below d_c .*

Furthermore, the diffusion constant of the charge carriers $D(T)$ can be evaluated from the mobility $\mu(T)$ using Eq. (8), as shown in Fig. 17. In Fig. 17, the diffusion constant can be reproduced using the relation

$$D(T) = D_0 T \exp\left(-\frac{\tilde{U}}{T - T_0}\right), \quad (25)$$

where D_0 is a positive constant. The value of D changes from 10^{-13} to 10^{-8} cm²/s as the temperature changes from 363 K to 410 K, for thin films of the amorphous polyamide copolymer PA66/6I with a thickness larger than d_c . Below d_c , an intrinsic increase in D is observed, similar to that for the mobility μ . Therefore, the translational motion of the charge carriers is intrinsically enhanced with decreasing film thickness.

V. DISCUSSION

From the observed dielectric spectra for thin films of PA66/6I, we successfully discerned the separate contributions from the EP process and the α process. The physical origin of the α process is due to segmental motion, which can be regarded as a rotational motion of polymeric segments, while that of the EP process is due to the translational motion

of charge carriers. Because the motion of charge carriers is activated via the molecular motion of the polyamide systems in which the charge carriers are located, the motion of charge carriers is strongly associated with the molecular motions of polymer chains.

A. Glassy dynamics evaluated from the EP process

As shown in Fig. 11, the temperature dependence of $\tau_{ep}(T)$ is well reproduced by the VFT law. Hence, we expect that the glass transition behavior of PA66/6I can be evaluated from the EP process. Figure 12 clearly shows that T_g is well evaluated using $\tau_{ep}(T)$. The fragility index evaluated from τ_{ep} increases with decreasing film thickness, as shown in Fig. 13. The dependence of T_g and the fragility index m on the film thickness has been measured for the thin films and/or nanoparticles of several polymers without strong polarity [41–43]. Both T_g and the fragility index decrease with decreasing film thickness for many cases. In contrast, the present result clearly shows that as the film thickness decreases, the fragility index increases, while T_g decreases. This inverse dependence on thickness of m might be associated with the strong polarity of PA66/6I. The fragility index can be regarded as a measure of the cooperativity of the α process or segmental motion [40]. In the case of polyamides, the formation of a network as a result of the hydrogen bonding can promote the cooperativity of the glassy dynamics, even in thin films.

B. Coupling or decoupling of translational and rotational motions

As shown in Fig. 15(a), the relaxation rate of the EP process shows an intrinsic increase below a critical thickness; that is, f_{ep} increases with decreasing film thickness with a stronger thickness dependence than that described by the linear relationship between f_{ep} and d^{-1} . This corresponds to the enhancement in the diffusion motion of the charge carriers in the thin films with $d < d_c$, as shown in Fig. 17. On the other hand, the relaxation rate of the α process shows almost no systematic dependence on the film thickness, for the thickness range investigated in this study, although there is scatter in the data points. This might be experimental evidence for the decoupling of the rotational motion of polymer chain segments and the translational motion of charge carriers in amorphous polyamide copolymers. More precise measurements are required in order to discuss this decoupling of the rotational motion and the translational motion in greater detail.

C. Origin of charge carriers

The dielectric relaxation spectra of amorphous polyamide copolymers clearly show strong signals for the EP process. As mentioned in Sec. I, there are several possible candidates for charge carriers in polyamides. One of the most promising consists of ionized protons. From the present analysis of the EP process on the basis of the theoretical model developed by Coelho [25], the diffusion constant and equilibrium concentration of the charge carriers are obtained as shown in Figs. 16 and 17. The charge carrier concentration $n_0(T)$ shows almost no significant temperature dependence for the temperature range

investigated in this study and has a value of $n_0 \approx 10^{18} \text{ cm}^{-3}$. The concentration of hydrogen atoms in the polyamide can be evaluated as $n_0^H \approx 5 \times 10^{21} \text{ cm}^{-3}$. Therefore, the ratio of the concentration of the charge carriers to that of the hydrogen atoms in the polyamide can be calculated as follows: $n_0/n_0^H = 0.02\%$. Hence, ionized protons are possible candidates for the charge carrier in PA66/6I.

The absolute values of the diffusion constants of the charge carriers obtained in this study can be compared with those of various particles in the literature in the following manner.

(1) It is well known that there is a possibility of water uptake in polyamide. The diffusion constant of adsorbed water molecules in amorphous polyamide copolymers has been measured by neutron scattering and is estimated to be between 10^{-6} and $10^{-5} \text{ cm}^2/\text{s}$ for the temperature range from 350 K to 430 K [20].

(2) The mutual diffusion constant of polystyrene (PS) chains at the interface between h-PS and d-PS has been reported in Ref. [44]. For $M_w = 10^5$, the mutual diffusion constant of PS changes from 2.0×10^{-16} to $1.2 \times 10^{-14} \text{ cm}^2/\text{s}$ as the temperature increases from 396 K to 413 K.

(3) Similar measurements of the diffusion constant of deuterated polyethylene (PE) in a matrix of hydrogenated PE show that the diffusion constant at 449 K is equal to $0.2 \times M_w^{-2.0}$ [45]. Here, for $M_w = 10^4$, the value of D is equal to $2 \times 10^{-9} \text{ cm}^2/\text{s}$.

(4) The tracer diffusion constant of the chloride ion Cl^- in polyamide-6 film has been measured in Ref. [46]. The diffusion constant ranges from 1.4×10^{-9} to $3.4 \times 10^{-8} \text{ cm}^2/\text{s}$, depending on the pH at room temperature.

(5) The calculated diffusion constant of the hydrogen combined to the terminal amines in polyamide-6,6 is reported to be $6 \times 10^{-13} \text{ cm}^2/\text{s}$ at 298 K [47].

The diffusion motion of water adsorbed on PA66/6I is much faster than the observed diffusion motion in this study. The observed diffusion constants of the chloride in polyamide-6 and the calculated ones of the combined hydrogens in polyamide-6,6 are larger than those observed in PA66/6I. On the other hand, the diffusion constants related to the reptation motion of PE chains are comparable to the observed values.

The comparison of the observed diffusion constants for PA66/6I in Fig. 17 with various diffusion constants in the literature suggests that the molecular motion of the charge carriers is strongly correlated with the motion of entire chains of the polyamide, if ionized protons are the actual charge carriers responsible for the EP process in PA66/6I. The observed results are quite consistent with this idea, although further direct measurements of both the molecular motion of ionized protons and the reptation motion of the polyamide are required for further investigation.

VI. CONCLUDING REMARKS

In this study, the dynamics of the electrode polarization process (EP process) and the α process in thin films of an amorphous polyamide copolymer were investigated using DRS measurements. The obtained results are summarized as follows.

(1) The relaxation time of the EP process has a VFT type of temperature dependence, and the T_g evaluated from the EP process agrees very well with the T_g determined from DSC measurements. The fragility index derived from the EP process increases with decreasing film thickness.

(2) There is a distinct deviation from this linear law for thicknesses smaller than a critical value. This deviation corresponds to an increase in the diffusion constant of the charge carriers, which are responsible for the EP process. The α process is located in a region of higher frequency than the EP process at high temperatures, but merges with the EP process near T_g .

(3) The dependence of the relaxation time of the α process on the film thickness is different from that of the EP process. This suggests that there is decoupling between the segmental motion of the polymers and the translational motion of charge carriers in confinement.

In the present study, we successfully derived the physical properties of the glassy dynamics from those of the EP process. This means that it is possible to evaluate the glassy dynamics even if the α process cannot be observed as a result of the existence of a large signal related to the motion of charges. Molecular motion, such as the α process, may be clarified in large polar polymeric systems through the EP process. For crystalline polyamide systems, the MWS interfacial polarization process should appear, in addition to the EP process. Even in such systems, the evaluation of the glassy dynamics from the EP process should be possible, in principle, although actual data analysis would be very difficult. We also showed that there is possible decoupling between the rotational motion of polyamide chain segments and the translational motion of charge carriers.

ACKNOWLEDGMENTS

This work was supported by a Grant-in-Aid for Scientific Research (B) (Grant No. 25287108) and Exploratory Research (Grant No. 25610127) from the **Japan Society for the Promotion of Science**. The authors would like to express their cordial thanks to Solvay for supplying them with the amorphous random copolymer PA66/6I. They also thank an anonymous reviewer for the very useful comments and suggestions. The synchrotron radiation experiments were performed at the BL40B2 of SPring-8 with the approval of the **Japan Synchrotron Radiation Research Institute** (JASRI) (Proposals No. 2013A1173 and No. 2014A1230).

- [1] K. L. Ngai, in *Physical Properties of Polymers*, edited by J. Mark (Cambridge University Press, Cambridge, UK, 2004), pp. 72–152.
- [2] G. Adam and J. H. Gibbs, *J. Chem. Phys.* **43**, 139 (1965).

- [3] L. Berthier, G. Biroli, J.-P. Bouchaud, L. Cipelletti, and W. van Saarloos, *Dynamical Heterogeneities in Glasses, Colloids, and Granular Media* (Oxford University Press, New York, 2011).
- [4] M. Alcoutlabi and G. B. McKenna, *J. Phys.: Condens. Matter* **17**, R461 (2005).

- [5] M. Tress, M. Erber, E. U. Mapesa, H. Huth, J. Müller, A. Serghei, C. Schick, K.-J. Eichhorn, B. Voit, and F. Kremer, *Macromolecules* **43**, 9937 (2010).
- [6] M. Y. Efremov, A. V. Kiyanova, J. Last, S. S. Soofi, C. Thode, and P. F. Nealey, *Phys. Rev. E* **86**, 021501 (2012).
- [7] I. Chang, F. Fujara, B. Geil, G. Heuberger, T. Mangel, and H. Sillescu, *J. Non-Cryst. Solids* **172–174**, 248 (1994).
- [8] M. D. Ediger, *Annu. Rev. Phys. Chem.* **51**, 99 (2000).
- [9] S. Napolitano, M. Wübbenhorst, and F. Kremer, editors, *Dynamics in Geometrical Confinement* (Springer, Berlin, 2014), pp. 279–306.
- [10] K. Fukao and Y. Miyamoto, *Phys. Rev. E* **61**, 1743 (2000).
- [11] S. Napolitano, S. Capponi, and B. Vanroy, *Eur. Phys. J. E* **36**, 61 (2013).
- [12] D. Labahn, R. Mix, and A. Schönhals, *Phys. Rev. E* **79**, 011801 (2009).
- [13] H. Yin, S. Napolitano, and A. Schönhals, *Macromolecules* **45**, 1652 (2012).
- [14] N. G. McCrum, B. E. Read, and G. Williams, *Anelastic and Dielectric Effects in Polymeric Solids* (Wiley, London, New York, 1967).
- [15] W. O. Baker and W. A. Yager, *J. Am. Chem. Soc.* **64**, 2171 (1942).
- [16] R. H. Boyd, *J. Chem. Phys.* **30**, 1276 (1959).
- [17] E. Laredo and M. C. Hernandez, *J. Polym. Sci., Part B: Polym. Phys.* **35**, 2879 (1997).
- [18] R. M. Neagu, E. Neagu, A. Kyritsis, and P. Pissis, *J. Phys. D: Appl. Phys.* **33**, 1921 (2000).
- [19] H. Lu, X. Zhang, B. He, and H. Zhang, *J. Appl. Polym. Sci.* **102**, 3590 (2006).
- [20] M. Laurati, P. Sotta, D. R. Long, L.-A. Fillot, A. Arbe, A. Alegría, J. P. Embs, T. Unruh, G. J. Schneider, and J. Colmenero, *Macromolecules* **45**, 1676 (2012).
- [21] A. Schönhals and F. Kremer, in *Broadband Dielectric Spectroscopy*, edited by F. Kremer and A. Schönhals (Springer, Berlin, Heidelberg, 2003), pp. 59–98.
- [22] H. Kiho, Y. Miyamoto, and H. Miyaji, *Polymer* **27**, 1542 (1986).
- [23] G. Blum, F. Kremer, T. Juworek, and G. Wegner, *Adv. Mater.* **7**, 1017 (1995).
- [24] J. R. Macdonald, *Phys. Rev.* **92**, 4 (1953).
- [25] R. Coelho, *Rev. Phys. Appl. (Paris)* **18**, 137 (1983).
- [26] R. J. Klein, S. Zhang, S. Dou, B. H. Jones, R. H. Colby, and J. Runt, *J. Chem. Phys.* **124**, 144903 (2006).
- [27] A. Einstein, *Ann. Phys. (Leipzig)* **322**, 549 (1905).
- [28] R. Kubo, M. Toda, and N. Hashitsume, *Statistical Physics II: Nonequilibrium Statistical Mechanics* (Springer, Berlin, Heidelberg, 1991).
- [29] A. Serghei, M. Tress, J. R. Sangoro, and F. Kremer, *Phys. Rev. B* **80**, 184301 (2009).
- [30] J. R. Sangoro and F. Kremer, *Acc. Chem. Res.* **45**, 525 (2012).
- [31] J. R. Sangoro, C. Iacob, A. L. Agapov, Y. Wang, S. Berdzinski, H. Rexhausen, V. Strehmel, C. Friedrich, A. P. Sokolov, and F. Kremer, *Soft Matter* **10**, 3536 (2014).
- [32] Y. Wang, C.-N. Sun, F. Fan, J. R. Sangoro, M. B. Berman, S. G. Greenbaum, T. A. Zawodzinski, and A. P. Sokolov, *Phys. Rev. E* **87**, 042308 (2013).
- [33] S. Havriliak and S. Negami, *Polymer* **8**, 161 (1967).
- [34] H. Vogel, *Phys. Z.* **22**, 645 (1921).
- [35] G. S. Fulcher, *J. Am. Ceram. Soc.* **8**, 339 (1925).
- [36] G. S. Fulcher, *J. Am. Ceram. Soc.* **8**, 789 (1925).
- [37] G. Tammann and W. Hesse, *Z. Anorg. Allg. Chem.* **156**, 245 (1926).
- [38] D. Boese and F. Kremer, *Macromolecules* **23**, 829 (1990).
- [39] G. R. Strobl, *The Physics of Polymers: Concepts for Understanding Their Structures and Behavior*, 3rd ed. (Springer, Berlin, Heidelberg, 2007), p. 263.
- [40] R. Böhmer and C. A. Angell, *Phys. Rev. B* **45**, 10091 (1992).
- [41] K. Fukao and Y. Miyamoto, *Phys. Rev. E* **64**, 011803 (2001).
- [42] C. M. Evans, H. Deng, W. F. Jager, and J. M. Torkelson, *Macromolecules* **46**, 6091 (2013).
- [43] C. Zhang and R. D. Priestley, *Soft Matter* **9**, 7076 (2013).
- [44] S. J. Whitlow and R. P. Wool, *Macromolecules* **24**, 5926 (1991).
- [45] J. Klein, *Nature (London)* **271**, 143 (1978).
- [46] G. Chantrey and I. D. Rattee, *J. Soc. Dyers Colour.* **85**, 618 (1969).
- [47] G. E. Zaikov, A. L. Iordanskii, and V. S. Markin, *Diffusion of Electrolytes in Polymers* (CRC Press, Boca Raton, FL, 1988).